Wood/Polymer Nanocomposites Prepared by Impregnation with Furfuryl Alcohol and Nano-SiO₂

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Wood/polymer nanocomposites were prepared by vacuum impregnation of furfuryl alcohol (FA) and nano-SiO₂ into fast-growing poplar wood. The nano-SiO₂ was mixed with FA solution, followed by *in-situ* polymerization of FA. The properties of nanocomposites and the effects of nanoparticles on these properties were investigated. Wood physico-mechanical properties, such as dimensional stability, density, water uptake, and surface hardness, were significantly improved. Moreover, the addition of nano-SiO₂ improved the surface hardness and dimensional stability of wood and kept the excellent properties of FA-treated wood. Thermogravimetric analysis indicated that the effect of nano-SiO₂ on thermostability was hindered. X-ray photoelectron spectroscopy and scanning electron microscopy with energy-dispersive X-ray spectroscopy showed that nano-SiO2 was successfully incorporated into wood via the action of FA, and diffused into the wood lumen and cell wall. X-ray diffraction results indicated the weakening of the crystallinity in the treated wood was due to the polymerization of FA.

*Keywords: Renewable resources; Wood/polymer nanocomposite (WPNC); Wood modification; Furfurylation; Nano-SiO*₂

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INTRODUCTION

Environmental and economic concerns have opened new opportunities for the development of wood modification. Many processes have been published in this field (Rowell 2005). The widely studied processes include acetylation, furfurylation, impregnation with lower molecular weight resins, cross-linking modification, and thermal modification (Deka and Saikia 2000; Lande *et al.* 2008; Esteves and Pereira 2009; Li *et al.* 2009; Cetin *et al.* 2011).

The application of biomass-derived substances for wood modification has the potential to reduce the depletion of petroleum resources. Among the biomass-derived chemicals, furfuryl alcohol (FA) derived from renewable saccharidic resources is an attractive intermediate for various useful polymer products (Moreau *et al.* 2004; Kim *et al.* 2014). With many advantages and development prospects, wood furfurylation becomes an excellent method in wood modification. This process is based on wood impregnation with FA and other agents, followed by *in-situ* polymerization at an elevated temperature. After decades of development, many properties of furfurylated wood have been studied, such as dimensional stabilization (Stamm 1977; Baysal *et al.* 2004), physico-mechanical properties (Schneider *et al.* 2000), moisture content (Thygesen and

Elder 2009), durability (De Vetter *et al.* 2009), and ecotoxicity (Lande *et al.* 2004). These studies show that furfurylated wood is an excellent and eco-friendly material, and consequently, it has been successfully commercialized. The furfurylated wood has been applied to construction materials, furniture, and flooring products (Lande *et al.* 2008). Moreover, many studies have focused on the mechanisms of FA polymerization (Choura *et al.* 1996; Nordstierna *et al.* 2008). Although many properties of wood can be improved by furfurylation, the enhancement of mechanical strength is not notable with a low weight percent gain (WPG), especially the modulus of elasticity (MOE), impact strength, and bending strength, which are important in wood applications. Moreover, the wood becomes darker after furfurylation (Herold *et al.* 2013), which also limits the application of this process. Hence, it is significant to investigate how to improve the wood strength with low WPG and alleviate discoloration, especially in fast-growing wood.

Nano-based treatments applied to wood modification present new opportunities to enhance wood properties. Wood polymer nanocomposites (WPNC) can be prepared with a range of improved physical, mechanical, and other properties. Many attempts have been made in this field in the past decades. A WPNC was prepared using water-soluble phenolic resin (PF) and montmorillonite (MMT) with improved surface mechanical properties, flame retardance, and dimensional stability (Xue and Zhao 2008). Solid aspen wood properties were improved, including surface hardness, MOE, dimensional stability, and water resistance, by impregnating wood with melamine-urea-formaldehyde (MUF) in combination with MMT (Cai *et al.* 2008). The impregnation of wood with urea formaldehyde (UF) and nano-SiO₂ improved the dimensional stability, flame resistance, and hardness (Shi *et al.* 2007). A WPC was prepared on the basis of nano-SiO₂ and nanoclay by impregnation of melamine formaldehyde-FA copolymer, and many properties were improved (Hazarika and Maji 2013).

Nano-SiO₂ is widely used in polymer composites to improve the mechanical strength and can be also applied to improve the properties of FA-modified wood. These organic/inorganic hybrid materials combine the properties of organic polymers and inorganic ceramics. In poly-furfuryl alcohol (PFA)/SiO₂ hybrid material, the nano-SiO₂ can restrict the motions of the furan chains and thus improve the internal rigidity and thermal stability of the organic/inorganic system (Guigo *et al.* 2009). Therefore, the PFA/SiO₂ hybrid material endows the wood with high mechanical strength and thermal stability.

With this in mind, we evaluated the feasibility of preparing wood polymer nanocomposites from furfurylated wood with nano-SiO₂. In addition, the location of PFA and nano-SiO₂ in the wood and the effects of nano-SiO₂ on properties of the wood were also investigated.

EXPERIMENTAL

Materials

Defect-free and straight-grained sapwood of fast-growing poplar wood (*Populus* spp.) was obtained from Guangxi Gaofeng Forest Farm, China. End-matched samples for both the modified and control samples were manufactured with a dimension of 20 mm \times 20 mm \times 20 mm. To reduce deviation, all samples were chosen from the same timber and machined (sawn and planed) at the same time.

Furfuryl alcohol (chemical purity 98.0%; Sinopharm Chemical Reagent Co., Ltd., China), nano-SiO₂ (particle diameter: 15 ± 5 nm; Shanghai Maikun Chemical Co., Ltd., China), maleic anhydride, and disodium tetraborate (analytically pure, Beijing Chemical Works, China) were used in these studies.

Methods

Preparation of impregnation solutions

Two types of impregnation solutions were prepared: 50 wt% FA water solution; and 50 wt% FA water solution containing 1.0 wt% nano-SiO₂ (mixed by sonication for 30 min). These dispersions were used for impregnation after the addition of 2 wt% maleic anhydride and 4 wt% disodium tetraborate.

Furfurylation

Samples were impregnated with FA and nano-SiO₂ *via* the following procedures. Drying: The samples were initially dried at 105 °C until reaching constant weight. Afterwards, the dimensions and weights were measured.

Impregnation: The samples were immersed in the solution using a vacuum drying oven (0.1 bar vacuum, 60 min). After impregnation, the excessive solution from the samples was blotted with tissue paper.

Polymerization: The samples were wrapped with aluminum foil, kept at room temperature for 12 h, and placed into an oven at 100 °C for another 12 h for polymerization. Afterwards, the aluminum foil was removed and the sample was dried at 105 °C until constant weight was obtained. The dimensions and weights were then measured. The WPG and 24 h deionized water immersion leachability (*L*) were calculated as follows:

WPG (%) =
$$(W_1 - W_0)/W_0 \times 100$$
 (1)

$$L(\%) = (W_1 - W_3)/(W_1 - W_0) \times 100$$
⁽²⁾

where W_0 and W_1 are the oven-dried weights of a specimen before treatment and after treatment, respectively, and W_3 is the oven-dried weights of treated samples after 24 h of water immersion.

X-ray photoelectron spectroscopy (XPS) analysis

The presence of silicon deposits in the samples were investigated by XPS (K-Alpha Thermo Scientific, USA) with a monochromatic Al K α X-ray source at a pressure of 10⁻⁸ mbar. Survey spectra were recorded with 1.0-eV steps and 200-eV analyzer pass energy, and high resolution regions were investigated with 0.1-eV steps and 50-eV pass energy. The spot size was roughly 400 μ m². All binding energies were calibrated with the C1s peak at 285 eV as a reference, which corresponds to the carbons in aromatic hydrocarbons. The ratio of elemental silica to carbon (Si/C) was determined from the low-resolution spectra.

Scanning electron microscopy (SEM) and energy-dispersive X-ray spectroscopy (EDX)

The penetration ability of PFA resin with nano-SiO₂ into the cell wall and its distribution within the cell wall were analyzed using SEM (Hitachi S-3400N, Japan). Therefore, tangential and radial sections of treated and untreated wood samples were mounted on conductive adhesives, sputter-coated with gold, and then observed under SEM at an accelerating voltage of 15 kV. In addition, the wood samples were also analyzed by SEM (Hitachi S-4800, Japan) coupled with energy dispersive X-ray analysis (EDX).

Fourier transform infrared spectrometry (FT-IR)

The treated and untreated wood samples were milled to a 200-mesh particle size and embedded into potassium bromide (KBr) pellets at a ratio of 1:100 in weight. The pellets were analyzed by FT-IR (Nicolet 6700 Thermo Scientific, USA) with a scanning range of 4000 to 400 cm⁻¹ at a resolution of 4 cm⁻¹ for 32 scans.

X-ray diffraction (XRD) analysis

The crystallinity of the treated and untreated wood samples was evaluated by XRD, using an XRD 6000 diffractrometer (Shimadzu, Japan). The apparatus parameters were as follows: Cu K α radiation with graphite monochromator, voltage 40 kV, electric current 30 mA and 2θ scan range of 5 to 40° with a scanning speed of 2° min⁻¹.

Measurement of physical and mechanical properties

Dimensional stability was evaluated by measurement of anti-swelling efficiency (ASE) as follows:

ASE (%) =
$$(S_u - S_t) / S_u \times 100$$
 (3)

where S_u is the volumetric swelling of an untreated sample and S_t is that of a treated sample.

Water uptake (WU) after 24 h of deionized water immersion and bulking effect (BE) were calculated as follows:

WU (%) =
$$(W_2 - W_1)/W_1 \times 100$$
 (4)

BE (%) =
$$(V_1 - V_0)/V_0 \times 100$$
 (5)

where W_2 is the weight of a wood sample after 24 h water immersion, and V_0 and V_1 are the dried wood volumes before and after treatment, respectively.

Surface hardness was measured with a TH210 durometer (Beijing TIME High Technology Ltd., China) and expressed as Shore D hardness according to ASTM D2240 (2010). Each type of sample was measured by 20 tests.

Thermogravimetric Analysis (TGA)

Thermal stabilities of the untreated and treated wood samples were evaluated by TGA, using a Q5000 TGA analyzer (TA Instruments, USA) at a constant heating rate of 20 °C min⁻¹ from room temperature to 600 °C under a flowing nitrogen atmosphere.

RESULTS AND DISCUSSION

After treatment, the WPNC was prepared with an even color distribution (*i.e.*, dark brown) from the exterior to the interior in the longitudinal section. Figure 1 shows the exterior and the interior of the untreated or treated wood with FA and FA/SiO₂. The average WPGs is 73.98% for the FA-treated wood and 76.47% for the FA/SiO₂-treated wood. The dark color is generated by the conjugated structure derived from the aliphatically linked PFA (Choura *et al.* 1996). Although the WPG of the FA/SiO₂-treated wood was higher than the FA-treated wood, its color was lighter, which is likely due to the conjugation confinement of nano-SiO₂ network in the PFA. The average leachabilities of the FA/SiO₂-treated samples were both less than 6.2%.



Fig. 1. Color of untreated and treated wood samples. Inside color: (a₁) untreated wood; (b₁) FA-treated wood; and (c₁) FA/SiO₂-treated wood. Surface color: (a₂) untreated wood; (b₂) FA- treated wood; and (c₂) FA/SiO₂-treated wood



Fig. 2. Survey XPS spectra of FA/SiO₂-treated and FA-treated samples

XPS Analysis

Figure 2 shows the typical XPS survey spectra for the treated wood samples. Inspection of these XPS spectra revealed that the C and Si occurred at 285 eV and 102 eV, respectively. For the FA/SiO₂-treated wood, the absolute differences in mol% were 97.8 and 2.2 At% (atom percentage) for C and Si, respectively, and the Si/C ratio was 0.0225. In comparison, the absolute differences of FA-treated wood were 98.38 and 1.61 At% for C and Si, respectively, and the Si/C ratio was 0.0164. The very small Si content of natural poplar wood concurs with the results of a previous study (Liao *et al.* 2004). Obviously, the Si content in the FA/SiO₂-treated wood was higher when compared with the FA-treated wood, which indicated the successful incorporation of nano-SiO₂ into the wood.

SEM Analysis

Figure 3 shows the morphologies of the untreated wood and the wood treated with FA or FA/SiO₂ at different magnifications. Empty cell walls, pits, and distinct lacunae were obvious in the untreated wood (Fig. 3a), while these empty places in the treated wood were occupied by the resin or nano-SiO₂ materials. Noticeably, the wood vessel walls were rough and the pits were filled with polymer, with many distinct wavy structures on the walls (Fig. 3b). These changes were due to the polymerization of FA. However, no notable polymer aggregation occurred on the surface, probably because the morphology of PFA is glasslike, and the polymer is well-distributed in the wood cell walls (Fitzer *et al.* 1969). It was deduced from these observations that the PFA was in intimate contact with the wood cell walls.



Fig. 3. Scanning electron micrographs: (a) untreated; (b) FA-treated; and (c) and (d) FA/SiO₂-treated wood samples

With the addition of nano-SiO₂ into the FA solution, many clusters adhered to the cell walls and even plugged the pits of the treated wood (Figs. 3c and 3d). With the presence of nanopores, wood can accommodate nanoparticles (Xue and Zhao 2008). Hence, these clusters could be a part of nano-SiO₂ particles that cannot enter the wood cell walls due to its agglomeration and separate from PFA. The diameter of these inorganic clusters was about 5 μ m. These observations supported the XPS data. In addition, a part of the nanoparticles was covered by PFA resin (Fig. 3c), indicating that the nano-SiO₂ can be incorporated with FA to form a composite material. In general, the effect of the nanoparticles on wood properties may be noticeable.

By SEM-EDX, the elemental composition of the FA- and FA/SiO₂-treated wood samples was determined, and the results are shown in Table 1. For the data of point 2, it can be determined that the agglomerates were comprised of SiO₂. Comparing the data between point 1 and 3, the distinct difference observed was a unique silicon signal in FA/SiO₂-treated wood sample. But the amount of silicon was low, which was coincident with the results of XPS. Also, little amounts of sodium were determined in both FA- and FA/SiO₂-treated wood samples, which was due to the disodium tetraborate in the FA solutions.

Table	1. SEM-EDX Va	lues of Carbon	, Oxygen,	and Silicon i	in Points 1, 2	, and 3
in Fig.	3b and c					

Point	Carbon (wt%)	Oxygen (wt%)	Silicon (wt%)
1	52.97	47.03	0
2	53.83	35.77	10.40
3	51.58	46.92	0.79

FT-IR Analysis

Figure 4 shows the FT-IR spectra of the four samples over the 400 to 4000 cm⁻¹ wavenumber range. For the wood samples, the prominent band at 3400 cm⁻¹ is attributed to the stretching of the -OH, the band at 1730 cm⁻¹ is assigned to C=O stretching of the acetyl group, the bands at 1596 and 1507 cm⁻¹ are assigned to the aromatic skeletal vibration of lignin, and the band at 1261 cm⁻¹ is assigned to C-O stretching of the guaiacyl ring (Pandey 1999). Two notable peaks at 1716 and 1562 cm⁻¹ only appeared in the treated woods, which are attributed to the C=O stretching vibration of the γ -diketone formed from hydrolytic ring opening of the furan of PFA and the skeletal vibration of 2,5-disubstituted furan rings, respectively (Chuang et al. 1984; Pranger and Tannenbaum 2008; Ahmad et al. 2013). This result also indicated the polymerization of the FA monomer in the wood. Two small bands at 1652 and 899 cm⁻¹ in curve b are assigned to the conjugated carbonyl stretching and the β -glucosidic linkages between the sugar units, respectively, whereas they disappeared in both curve c and d (Kačuráková et al. 1998; Sun et al. 2001). This phenomenon may be attributed to the acid hydrolysis of lignin and hemicellulose in the poplar wood (Lee and McCaskey 1983; Sun et al. 2013). The absorption bands in the spectra of two treated woods were not different due to the low amount of SiO₂ particle impregnation, which indicated that the FA was successfully polymerized in the presence of nano-SiO₂ and that no chemical interaction occurred between the SiO_2 and the FA or between the SiO_2 and the wood.



Fig. 4. FT-IR spectra: (a) nano-SiO₂; (b) untreated wood; (c) FA-treated wood; and (d) FA/SiO₂-treated wood

XRD Analysis

Figure 5 shows the XRD spectra of untreated-wood, as well as wood treated with water, FA, or FA/SiO₂. For both untreated and water-treated woods, the XRD peaks mainly occurred at diffraction angles of 16.8°, 22.2°, and 34.6°, and are attributed to the cellulose crystal planes I₁₀₁, I₀₀₂, and I₀₄₀, respectively. Moreover, an obvious trough, which is a characteristic of the amorphous wood region, appeared at 18.2° (Cave 1997). However, the patterns of FA- and FA/SiO₂-treated woods were substantially changed, with the trough at 18.2° obviously elevated. There was no noticeable difference in the diffractogram of the FA-treated wood with the addition of nano-SiO₂ particles. The difference of peaks at 37.8° may be due to some signal extraneous to the wood composition, and it had little impact on the results. According to a previous report by Zavadskii (2004), the crystallinity of untreated wood was 37.75%, which increased to 46.30% after the wood's treatment with water. In the present furfurylation after 12 h of cure time, the hot water treatment led to rearranged or reoriented cellulose molecules or the crystallization in quasicrystalline of the amorphous region (Bhuiyan et al. 2000; Howell et al. 2011). The same phenomenon occurred in the FA- or FA/SiO₂-treated wood. Specifically, the crystallinities of FA- and FA/SiO₂-treated woods were 42.29% and 43.83% respectively, which indicated that the addition of SiO₂ into the wood slightly increased the crystallinity. The decreased crystallinity in the FA-treated wood when compared to the water-treated wood may be attributed to the amorphous structure of PFA. The FA monomer entered the amorphous region of the cellulose and polymerized in situ, which resulted in the swelling of the amorphous region. From this observation, the efficiency of furfurylation may be mainly a filling action, though the grafting between PFA and wood revealed using some model compounds could increase the crystallinity (Nordstierna et al. 2008; Wu et al. 2010).



Fig. 5. XRD spectra: (a) untreated wood; (b) water-treated wood; (c) FA-treated wood; and (d) FA/SiO₂-treated wood

Evaluation of Physical and Mechanical Properties

Table 2 shows the results for the bulk effect (BE), the water uptake (WU), the anti-swelling efficiency (ASE), and the surface hardness of the untreated and treated wood samples. Compared to the untreated wood, the BE, ASE, and surface hardness of the treated wood samples were augmented, while the WU was reduced by about 60%, which was attributed to the over bulking as well as the obstruction of cell wall pits by the hydrophobic PFA. In the FA/SiO₂-treated wood, the addition of nano-SiO₂ also slightly affected the dimensional stability and surface hardness, suggests that the silica cage restrict the segmental motion of furan chains *via* nanoconfinement and this organic/inorganic hybrid material improves the internal rigidity of the nanocomposite (Guigo *et al.* 2009).

Table 2. Physical and Mechanical Properties of Untreated and Treated Woo	d
Samples	

Wood sample	BEª	WUª	ASEª	Hardness [▶]				
	(%)	(%)	(%)	(Shore D)				
Untreated*	2.99 (±0.40)	113.32 (±7.88)	0 (±2.99)	46.97 (±4.25)				
FA-treated	15.02 (±1.08)	44.55 (±5.55)	65.75 (±0.86)	63.14 (±4.35)				
FA/SiO ₂ -treated	15.38 (±0.33)	41.38 (±3.68)	68.67 (±1.72)	69.28 (±3.45)				
^a Each value represents the average of six samples ± standard deviation								
^b Each value represents the average of twenty samples								
*Treated with water								

Evaluation of Thermal Stability

Thermal stability of the untreated and treated wood samples was studied using TGA. Figure 6 shows the mass loss of these samples as a function of temperature. The thermal degradation profiles revealed that most of the degradation events occurred between 200 and 400 °C. For the untreated wood, the first degradation region from 50 to 120 °C is related to the release of moisture and adsorbed water from the wood. The weight loss of the treated wood is lower at this temperature range, which implies that the water uptake of the wood was reduced after its modification. This finding also agrees with the previous results of the water uptake test. The untreated wood did not lose more than 20% of its weight before leveling off at 300 °C, whereas the weight of the treated wood distinctly declined from 200 to 300 °C, which was mainly attributed to the thermodegradation of PFA (Guigo *et al.* 2009).

The main pyrolysis process occurred at 366 °C in the untreated wood and at 333 °C in the FA-treated wood, and slightly higher in the FA/SiO₂-treated wood than the FA-treated wood. However, the nano-SiO₂ treatment did not noticeably affect the thermal stability of the composites, probably because the high WPG hindered the efficacy of the nano-SiO₂. The SEM micrographs showed that a large part of nano-SiO₂ may be separated from the PFA by the cell wall, and the synergistic effect of nano-SiO₂ and furanic matrix cannot work. This may be an alternative explanation for the observed result. Because of the high carbon content of PFA, the residual weight for the treated wood was almost two times higher than that of the untreated wood.



Fig. 6. TGA curves of untreated, FA-treated, and FA/SiO₂-treated wood

CONCLUSIONS

- 1. Wood-polymer nanocomposites (WPNC) were successfully prepared from fastgrowing poplar wood by impregnation using furfuryl alcohol (FA) and nano-SiO₂. This process of WPNC using FA and SiO₂ is feasible for the preparation of an effective product with improved properties.
- 2. SEM-EDX and XPS analyses showed that the nano-SiO₂ was incorporated into wood and diffused in the wood lumen and cell walls when using a FA solution as the medium. The FT-IR results indicated that successful polymerization of FA in wood occurred. However, there was no evidence found for the reaction among the SiO₂, wood, and FA from the FT-IR analysis.
- 3. The crystallinity of the FA-treated and FA/SiO₂-treated composites decreased relative to the untreated wood, as measured by XRD; this may be attributed to the penetration of PFA into the amorphous region of the wood cell wall. Wood properties such as surface hardness, water uptake, and anti-swelling efficiency were improved. The addition of nano-SiO₂ can enhance the dimensional stability and surface hardness of FA-treated wood. The TGA analysis showed that FA/SiO₂ treatment hindered the thermal stability of the wood composite. A more effective process for incorporating nano-SiO₂ and PFA into the wood substrate should be investigated.

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