# Influence of Residual Black Liquor in Pulp on Wastewater Pollution after Bleaching Process

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The influence of residual black liquor in pulp on wastewater pollution after the bleaching process was studied. The results show that the  $COD_{Cr}$  in bleaching effluent has a remarkable linearity with bleaching loss of pulps without residual black liquor. For pulps with some residual black liquor, more than 34% of the overall  $COD_{Cr}$  is produced by the residual black liquor. It follows that more effective washing to reduce the residual black liquor is an appropriate way to control the pollutant discharges from pulp and paper mill industry.

Keywords: Residual black liquor; Bleaching loss; COD<sub>Cr</sub>; Wastewater pollution

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#### INTRODUCTION

In terms of fresh water usage, the pulp and paper mill industry is a water-intensive industry; it is ranked third in the world after metals and chemical industries (Merayo *et al.* 2013). Wood preparation, pulping, bleaching, and coating operations are the main sources of pollution (Mahesh *et al.* 2016). Effluents of the pulp and paper industry contain a number of toxic compounds. The rejection of this effluent in nature without any treatment is responsible of serious damage for the environment and constitutes a threat for human health (Lara *et al.* 2003).

Kraft (or sulfate) and soda are the two major alkaline processes to produce chemical pulps (Cardoso *et al.* 2009). Cellulose fibers are disassociated from lignin by chemical reactions, which occur in a pressurized digester where wood chips or fibers are heated and cooked with the cooking liquor, composed basically of NaOH (sodium hydroxide), and in the case of the kraft process, also sodium sulfide (Cao *et al.* 2011; Zheng *et al.* 2012). The products from the digester reactions are fibers and black liquor. Black liquor is one of the main by-products of pulp paper industry, which is considered a pollutant because it contains about 50% of lignin (Zaied and Bellakhal 2009).

Some new methods for the treatment of black liquor have been researched, such as the oxidizing action of the Fenton reagent in the presence of solar UV irradiation (Fe<sup>2+</sup>/H<sub>2</sub>O<sub>2</sub>/UV) (Torrades *et al.* 2003), ultra-filtration for the recovery of lignin starting from black liquor (Wallberg *et al.* 2005; Wallberg and Jönsson 2006; Wallberg *et al.* 2003), and the electrocoagulation method (El-Ashtoukhy *et al.* 2009; Khansorthong and Hunsom 2009; Zaied and Bellakhal 2009). But none of them is widely used industrially due to high cost. At present, soda recovery units (Licursi *et al.* 2015) have been mainly applied for the treatment of black liquor. However, limitations of equipment lead to only about 85% recovery ratio for black liquor. This means that more than 15% of the black liquor remains in the pulp, and unfortunately it will be passed forward into the bleaching

process.

Bleaching of pulp using chlorine-based agents is still practiced in China and in other developing countries (Deshmukh et al. 2009). Due to the presence of chlorinated organic compounds, wastewater from bleaching units is toxic (Savant et al. 2006). These organochlorine compounds are collectively termed as adsorbable organic halides (AOX). AOX is a general parameter determining the total amount of organically bond halogens in wastewater. Many AOX compounds are persistent in the environment and show a significant toxic effect on human beings (Höfl et al. 1997; Barroca et al. 2001). About 500 different chlorinated organic compounds have been identified in paper mill effluent (Savant et al. 2006). Due to the universal harmful effect of organic halogens, many countries including China have a series of discharge standards for industrial wastewater (Xie *et al.* 2016). There are a number of approaches that have been explored to reduce the AOX level in paper mill effluent, which can be categorized as physic-chemical and biological (Kumar 2013). Physic-chemical approaches include adsorption, ultrafiltration, nanofiltration, and reverse osmosis (Patel and Suresh 2008). Biological techniques involve the use of diverse kinds of micro-organisms like bacteria, fungi, algae, and microbes of extreme habitats for reducing AOX and chromophores in pulp mill effluent (Belmonte et al. 2006; Ruggaber and Talley 2006; Morales et al. 2015). Although these methods of treatment seem effective for the wastewater of pulp industry, they present the disadvantage of being expensive because of their operating cost and the high cost for the chemical reagents used. How to reduce pollution emissions in the production process has become urgent for the pulp and paper industry.

There is a lack of literature about effects of remained black liquor entering the bleaching system on the  $COD_{Cr}$  of wastewater. This paper investigated the effects of residual black liquor in pulp on  $COD_{Cr}$  after the H-bleaching process, which is supposed to bring about fundamental changes in pollution control and treatments for pulping and paper-making industry in China.

### EXPERIMENTAL

#### **Raw Materials**

*Pinus koraiensis* was used as the raw material, with alkaline cooking. The total solids content of the black liquor was 194.3 g/L. The three-stage adverse current washing method was adopted to wash the black liquor out of the pulp at 60 °C to simulate industrial washing. A stock suspension having 35% (or alternatively 30%) consistency has 0.72 (or alternatively 3.71) g of black liquor residual solids in the pulp suspension. As shown in Table 1, four samples were selected according to different degrees of cleanliness for further analysis. The No. 1 pulp was thoroughly washed without any residual black liquor (ideal condition) and has a Kappa index of 27.58, whiteness of 34.32 % ISO, and viscosity of 474 mL  $\cdot$  g<sup>-1</sup>. There are different amounts of black liquor in the No. 2, No. 3, and No. 4.

Slurry	No. 1	No. 2	No. 3	No. 4
Dry pulp (g)	20	19.6	18.2	16.1
Black liquor (g)	0	2.4	10.8	23.52
Solids (g)	0	0.4	1.8	3.9

Table 1. Main	<b>Components</b>	and Cleanness	in 20 g Slur	ry of Different Pulps
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### **Bleaching Process of Pulp**

In the pulp bleaching section, the amount of pulp was 20 g on a dry solids basis. Prior to experiments, these samples had been equilibrated to 10%. The runs were carried out with different doses of active chlorine (4%, 7%, 10%, 13%, 16%) and the other parameters were kept fixed. Single stage hypochlorite bleaching was adopted at 52 °C for 25 min in a water bath. Mixing was accomplished by means of stirring every 10 min in order to make the bleaching reaction better; 8% sodium hydroxide was applied to adjust the pH, and the pH values of the bleaching solution were measured before the reaction, during bleaching, and after the reaction (Diel *et al.* 2016).

Bleaching effluent was collected and the pulp was washed thoroughly with enough water to remove suspended matter and particles for the future study.

### **Bleaching Process of Black Liquor**

To study the effects of residual black liquor on pollution loads in the bleaching process, the amount of black liquor of No. 1 (2, 3, 4) was 0 (2.4, 10.8, 23.5) g and the amount of solids of No. 1 (2, 3, 4) was 0 (0.4, 1.8, 3.9) g. The same bleaching method was applied to bleaching process of black liquor.

### **Other Analytical Method**

Prior to experiments, all liquid effluent samples were filtered with a membrane filter (pore size  $0.45 \ \mu m$ ) to remove suspended matter and particles. The determination of the COD<sub>Cr</sub> was carried out using a Hach spectrophotometer (DR2800, Hach, Loveland, CO, USA), according to standard methods (Karichappan *et al.* 2014). The pH was measured with a Sartorious PB-10 (Sartorius, Germany) pH meter.

### **RESULTS AND DISCUSSION**

# Kappa Index and Bleaching Loss of No. 1 and COD<sub>cr</sub> of No. 1 after Bleaching

The No. 1 pulp sample without residual black liquor was H-bleached at different amounts of available chlorine of 4%, 7%, 10%, 13%, and 16%, and Kappa index and bleaching loss of No. 1 and  $COD_{cr}$  of No. 1 after bleaching are shown in Table 2.

# **Table 2.** Kappa Index and Bleaching Loss of No. 1 and COD<sub>cr</sub> of No. 1 after Bleaching

Available Chlorine Content	4%	7%	10%	13%	16%
Kappa index	14.24	8.94	8.64	6.06	4.85
Bleaching loss (%)	3.00	4.80	5.70	6.10	7.15
CODcr (mg·L <sup>-1</sup> )	2823.1	4497.6	5274.5	5617.2	6150.0

As is shown in Table 2, with the increase of available chlorine, the kappa index of pulps decreased, indicating that lignin was disassociated from fibers. When the available chlorine was 16%, the removal rate of lignin was about 82.4%. The bleaching loss and  $COD_{cr}$  increased with the increase of the available chlorine. A linear relationship between bleaching loss and  $COD_{cr}$  was obtained, as is shown in Fig. 1.



**Fig. 1.** Relationship between bleaching loss and  $COD_{cr}$  in effluents for pulps without residual black liquor

For pulps without residual black liquor, the  $COD_{cr}$  in the wastewater are contributed to lignin and carbohydrates that are disassociated from the cellulose fibers (solid phase lignin). For the same fiber materials with the same bleaching conditions, the  $COD_{cr}$  contributed to solid phase lignin is calculated as  $COD_1$  in the next experiments.

# Kappa Index and Bleaching Loss of Pulp with Residual Black Liquor and COD<sub>cr</sub> of Pulp with Residual Black liquor after Bleaching

H-bleaching was applied to the No. 2, No. 3, and No. 4 pulp samples under different available chlorine dosages of 4%, 7%, 10%, 13%, and 16%. The kappa index and bleaching loss of No.2 (3, 4) and  $COD_{Cr}$  of No.2 (3, 4) are shown in Tables 3, 4 and 5, respectively. The  $COD_0$  is the actual determination values in bleaching effluents, which is composed of  $COD_1$  produced by solid phase lignin, and  $COD_L$  originated from the theoretical residual black liquor. According to Fig. 1,  $COD_1$  can be calculated with bleaching loss.

To test the reliability of the results obtained by calculating the  $COD_{L}$  from  $COD_{0}$  and  $COD_{1}$ , different amounts of black liquor were H-bleached in the same bleaching conditions, and the determination of  $COD_{Cr}$  in bleached residual black liquor was obtained as  $COD_{2}$ , as shown in Tables 3, 4, and 5.

The relative standard deviation (RSD) of the  $COD_L$  and  $COD_2$  ranged from 0.10% to 3.15%. Thus, the calculated values were in good agreement with the measured values, and it was feasible to calculate the  $COD_1$  by Fig. 1 to determine the pollution load from solid phase lignin. For all pulp samples, with increasing available chlorine content, the kappa index decreased gradually.

For the same slurry, the  $COD_L$  from residue black liquor had a greater effect on the wastewater pollution load than  $COD_1$  from solid phase lignin without No.2 pulp, but

 $COD_L$  from residue black liquor in the wastewater pollution load of No.2 was still evident. This is because the residual black liquor mainly includes the small molecule lignin compounds, hemicellulose, and some organic degradation of sugars after cooking (Sun *et al.* 1999; Mussatto *et al.* 2007). However, with the increase of available chlorine, the increase of the COD<sub>L</sub> is smaller than that of COD<sub>1</sub> in wastewater, because the amount of residue black liquor in the slurry is certain. Another possible reason is that aldehyde or alcohol organics in the black liquor are oxidized to the acid organics by adding one more oxygen, resulting in a smaller COD<sub>L</sub> change.

Available Chlorine Content	4%	7%	10%	13%	16%
Kappa index	16.64	12.82	10.42	9.28	7.57
Bleaching loss (%)	2.43	3.48	5.22	5.97	6.84
COD₀ (mg·L <sup>-1</sup> )	4360.5	5436	7506.6	8470	9182
COD₁ (mg·L <sup>-1</sup> )	2470.3	3334.1	4765.5	5382.5	6098.2
(COD <sub>1</sub> /COD <sub>0</sub> ) (%)	57	61	63	64	66
COD <sub>L</sub> (mg·L <sup>-1</sup> )	1890.2	2101.9	2741.1	3087.5	3083.8
COD <sub>2</sub> (mg·L <sup>-1</sup> )	1882.4	2238.6	2610.8	3147.5	3189.3
*CODL=COD0-COD1					

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Table 4. No. 3 Pulp Properties and Pollution Load after H-Bleaching

Available Chlorine Content	4%	7%	10%	13%	16%
Kappa index	18.57	16.62	14.13	12.29	11.72
Bleaching loss (%)	1.21	1.87	2.91	3.79	4.67
COD₀ (mg·L <sup>-1</sup> )	9828.5	10709	12263	12998	13852
COD₁ (mg·L⁻¹)	1465.6	2008.0	2866.9	3590.1	4313.33
(COD <sub>1</sub> /COD <sub>0</sub> ) (%)	15	19	23	28	31
COD <sub>L</sub> (mg·L <sup>-1</sup> )	8362.9	8701.0	9396.1	9407.9	9538.7
COD₂ (mg·L <sup>-1</sup> )	8209.4	8924.1	9262.0	9346.2	9621.8
*CODL=COD0-COD1					

Table 5. No. 4 Pulp F	Properties and Po	ollution Load after	H-Bleaching
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Available Chlorine Content	4%	7%	10%	13%	16%
Kappa index	22.27	18.94	17.27	15.45	12.12
Bleaching loss (%)	0.56	1.11	1.67	1.98	2.65
COD₀ (mg·L <sup>-1</sup> )	21062	23085	23837	24664	26706
COD₁ (mg·L⁻¹)	931.93	1385.3	1842.3	2096.2	2654.8
(COD <sub>1</sub> /COD <sub>0</sub> ) (%)	4.4	6.0	7.7	8.5	9.9
COD <sub>L</sub> (mg·L⁻¹)	20130	21700	21995	22568	24051
COD₂ (mg·L⁻¹)	20110	21550	21894	23013	24610
*COD <sub>L</sub> =COD <sub>0</sub> -COD <sub>1</sub>					

For different slurries at the same amount of available chlorine, with the increase of residual black liquor in the pulp,  $COD_0$  and  $COD_L$  gradually increased and bleaching losses and  $COD_1$  decreased. Thus, the liquid organics are more likely to react with bleaching agent than solid phase lignin; in a bleaching process for pulps with residual black liquor, most of bleaching agents react with the solubilized organics first. This reaction weakens the bleaching efficiency and wastes bleaching agents. A decrease in residual black liquor can reduce the dosage of bleaching agent, which may lessen the production of AOX and decrease the toxicity of wastewater (Rey *et al.* 2013; Kumar Chenna et al. 2016).

#### Contribution of Residual Black Liquor to CODcr in Bleaching Effluents

The contribution of residual black liquor to  $COD_{cr}$  after H-bleaching is shown in Fig. 2. As shown in Fig. 2 and Tables 3, 4, and 5, for different slurries in the same amount of available chlorine, a greater amount of residual black liquor resulted in a greater  $COD_{Cr}$ . For the same slurry, with the increase of available chlorine content, the contribution of residual black liquor to  $COD_{cr}$  in the wastewater decreased gradually. But all  $COD_{L}/COD_{0}$  ratio values were more than 34%. This result reveals that  $COD_{cr}$  from the residual black liquor was greater.



Fig. 2. Effect of black liquor on wastewater

### CONCLUSIONS

- 1. In the H-bleaching process, the relationship of bleaching loss and COD<sub>Cr</sub> in bleaching effluents is remarkably linear for pulps without residual black liquor.
- 2. For pulps with residual black liquor in H-bleaching process, most of bleaching agent reacts with residual black liquor in the pulp first. The results show that the COD<sub>Cr</sub> produced by residual black liquor accounted for more than 34% of the overall COD<sub>Cr</sub>.

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### **REFERENCES CITED**

- Barroca, M. J. M. C., Seco, I. M., Fernandes, P. M. M., Ferreira, L. M. G. A., and Castro, J. A. A. M. (2001). "Reduction of aox in the bleach plant of a pulp mill," *Environmental Science & Technology* 35(21), 4390-4393.
- Belmonte, M., Decap, J., Martínez, M., and Vidal, G. (2006). "Effect of aerobic sludge with increasing level of adaptation on abietic acid biodegradation," *Bulletin of Environmental Contamination and Toxicology* 77(6), 861-867. DOI: 10.1007/s00128-006-1224-3
- Cao, C., Guo, L., Chen, Y., Guo, S., and Lu, Y. (2011). "Hydrogen production from supercritical water gasification of alkaline wheat straw pulping black liquor in continuous flow system," *International Journal of Hydrogen Energy* 36(21), 13528-13535. DOI: 10.1016/j.ijhydene.2011.07.101
- Cardoso, M., Éder Domingos de Oliveira, and Passos, M. L. (2009). "Chemical composition and physical properties of black liquors and their effects on liquor recovery operation in brazilian pulp mills," *Fuel* 88(4), 756-763. DOI: 10.1016/j.fuel.2008.10.016
- Deshmukh, N.S., Lapsiya, K.L., and Savant, D.V. (2009). "Upflow anaerobic filter for the degradation of adsorbable organic halides (AOX) from bleach composite wastewater of pulp and paper industry," *Chemosphere* 75(9), 1179-85. DOI: 10.1016/j.chemosphere.2009.02.042
- Diel, C. L., Canevesi, R. L. S., Zempulski, D. A., Awadallak, J. A., Borba, C. E., Palú, F., and Silva, E. A. (2016). "Optimization of multiple-effect evaporation in the pulp and paper industry using response surface methodology," *Applied Thermal Engineering* 95, 18-23. DOI: 10.1016/j.applthermaleng.2015.10.136
- El-Ashtoukhy, E.S.Z., Amin, N.K., and Abdelwahab, O. (2009). "Treatment of paper mill effluents in a batch-stirred electrochemical tank reactor," *Chemical Engineering Journal* 146(2), 205-210. DOI: 10.1016/j.cej.2008.05.037
- Höfl, C., Sigl, G., and Specht. O. (1997). "Oxidative degradation of AOX and COD by different advanced oxidation processes: A comparative study with two samples of a pharmaceutical wastewater," *Water Science & Technology* 35(4), 257-264. DOI: 10.1016/S0273-1223(97)00033-4
- Karichappan, T., Venkatachalam, S., Jeganathan, P. M., and Karichappan, T. (2014). "Analysis of Efficiency of *Bacillus subtilis* to treat bagasse based paper and pulp industry wastewater – A novel approach," *Journal of the Korean Chemical Society* 58(2). DOI: 10.5012/jkcs.2014.58.2.198
- Khansorthong, S., and Hunsom, M. (2009). "Remediation of wastewater from pulp and paper mill industry by the electrochemical technique," *Chemical Engineering Journal* 151(1-3), 228-234. DOI: http://dx.doi.org/10.1016/j.cej.2009.02.038
- Kumar, R. (2013). "Biological AOX removal of pulp mill plant effluent by *Pseudomonas* aeruginosa – Bench study," Journal of Environmental Engineering & Landscape Management 21(4), 296-304. DOI: 10.3846/16486897.2012.745413
- Kumar Chenna, N., Piovano, P., Järnefelt, C., and Vuorinen, T. (2016). "Organochlorine formation in tertiary amine catalyzed pulp bleaching," *Journal of Wood Chemistry and Technology* 36(5), 318-328. DOI: 10.1080/02773813.2016.1148166
- Lara, M. A., RodríGuez-Malaver, A. J., Rojas, O. J., Holmquist, O., González, A. M., and Bullón, J. (2003). "Black liquor lignin biodegradation by *Trametes elegans*," *International Biodeterioration & Biodegradation* 52(3), 167-173. DOI:

10.1016/S0964-8305(03)00055-6

- Licursi, D., Antonetti, C., Bernardini, J., Cinelli, P., Coltelli, M. B., Lazzeri, A., Martinelli, M., and Galletti, A. M. R. (2015). "Characterization of the *Arundo donax* L. solid residue from hydrothermal conversion: Comparison with technical lignins and application perspectives," *Industrial Crops & Products* 76, 1008-1024.DOI: 10.1016/j.indcrop.2015.08.007
- Mahesh, S., Garg, K. K., Srivastava, V. C., Mishra, I. M., Prasad, B., and Mall, I. D. (2016). "Continuous electrocoagulation treatment of pulp and paper mill wastewater: Operating cost and sludge study," *RSC Advances* 6(20), 16223-16233. DOI: 10.1039/C5RA27486A

Merayo, N., Hermosilla, D., Blanco, L., Cortijo, L., and Blanco, Á. (2013). "Assessing the application of advanced oxidation processes, and their combination with biological treatment, to effluents from pulp and paper industry," *Journal of Hazardous Materials* 262(8), 420-427. DOI: 10.1016/j.jhazmat.2013.09.005

- Morales, G., Pesante, S., and Vidal, G. (2015). "Effects of black liquor shocks on activated sludge treatment of bleached kraft pulp mill wastewater," *Journal of Environmental Science & Health Part A Toxic/hazardous Substances & Environmental Engineering* 50(6), 639-645. DOI: 10.1080/10934529.2015.994974
- Mussatto, S. I., Fernandes, M., and Roberto, I. C. (2007). "Lignin recovery from brewer's spent grain black liquor," *Carbohydrate Polymers* 70(2), 218-223. DOI: 10.1016/j.carbpol.2007.03.021
- Patel, U. D., and Suresh, S. (2008). "Electrochemical treatment of pentachlorophenol in water and pulp bleaching effluent," *Separation & Purification Technology* 61(2), 115-122. DOI: 10.1016/j.seppur.2007.10.004
- Rey, M. D., Font, R., and Aracil, I. (2013). "Biogas from MSW landfill: Composition and determination of chlorine content with the AOX (adsorbable organically bound halogens) technique," *Energy* 63, 161-167. DOI: 10.1016/j.energy.2013.09.017
- Ruggaber, T. P., and Talley, J. W. (2006). "Enhancing bioremediation with enzymatic processes: a review," *Practice Periodical of Hazardous Toxic & Radioactive Waste Management* (10), 73-85. DOI: 10.1061/(ASCE)1090-025X(2006)10:2(73)
- Savant, D. V., Abdulrahman, R., and Ranade, D. R. (2006). "Anaerobic degradation of adsorbable organic halides (AOX) from pulp and paper industry wastewater," *Bioresource Technology* 97(9), 1092-1104. DOI: 10.1016/j.biortech.2004.12.013
- Sun, R., Tomkinson, J., and Bolton, J. (1999). "Effects of precipitation pH on the physico-chemical properties of the lignins isolated from the black liquor of oil palm empty fruit bunch fibre pulping," *Polymer Degradation and Stability* 63(2), 195-200. DOI: 10.1016/S0141-3910(98)00091-3
- Torrades, F., Pérez, M., Mansilla, H. D., and Peral, J. (2003). "Experimental design of Fenton and photo-Fenton reactions for the treatment of cellulose bleaching effluents," *Chemosphere* 53(10), 1211-20. DOI: 10.1016/S0045-6535(03)00579-4
- Wallberg, O., Holmqvist, A., and Jönsson, A.-S. (2005). "Ultrafiltration of kraft cooking liquors from a continuous cooking process," *Desalination* 180(1), 109-118. DOI: 10.1016/j.desal.2004.12.032
- Wallberg, O., and Jönsson, A.-S. (2006). "Separation of lignin in kraft cooking liquor from a continuous digester by ultrafiltration at temperatures above 100 C," *Desalination* 195(1), 187-200. DOI: 10.1016/j.desal.2005.11.011
- Wallberg, O., Jönsson, A. S., and Wimmerstedt, R. (2003). "Ultrafiltration of kraft black liquor with a ceramic membrane ☆," *Desalination* 156, 145-153. DOI:

10.1016/S0011-9164(03)00337-0

- Xie, Y., Chen, L., and Liu, R. (2016). "Oxidation of AOX and organic compounds in pharmaceutical wastewater in RSM-optimized-Fenton system," *Chemosphere* 155:217. DOI: 10.1016/j.chemosphere.2016.04.057
- Zaied, M., and Bellakhal, N. (2009). "Electrocoagulation treatment of black liquor from paper industry," *Journal of Hazardous Materials* 163(2-3), 995-1000. DOI: 10.1016/j.jhazmat.2008.07.115
- Zheng, Y., Chai, L. Y., Yang, Z. H., Tang, C. J., Chen, Y. H., and Shi, Y. (2012).
  "Enhanced remediation of black liquor by activated sludge bioaugmented with a novel exogenous microorganism culture," *Applied Microbiology & Biotechnology* 97(14), 6525-35. DOI: 10.1007/s00253-012-4453-x

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