

Chromones and Tannins from the Fruit of *Euscaphis japonica* var. *wupingensis*

Shunzhi Liu,^{a,b} Tao Zhu,^{a,c} Yingkun Qiu,^d Wenyu Qi,^a Hui Wu,^a Bangping Cai,^{a,b,*} and Jinguo Lin^{a,*}

Euscaphis japonica var. *wupingensis* is a variety of *E. japonica* in the family Staphyleaceae. The biological characteristics of *E. japonica* var. *wupingensis* have been well described, but few studies have been conducted on its constituents. After repeated separation using column chromatography and preparative high performance liquid chromatography (HPLC), ten compounds were isolated from the methanol extract of fruit of *E. japonica* var. *wupingensis*. The structure of the components was characterized on the basis of spectroscopic data, including nuclear magnetic resonance (NMR), and mass spectra (MS). These compounds included five chromone derivatives: isobiflorin (**1**), biflorin (**2**), 2-methyl-5,7-dihydroxy-chromone-7-O-β-D-glucopyranoside (**5**), 5,7-dihydroxy-2-methyl-4H-chromen-4-one (**6**), and quercetin-3-O-D-arabinoside (**7**), and four tannins: eugeniin (**3**), ellagic acid (**8**), 3, 3'-di-O-methylellagic acid 4-(5''-acetyl) - α -L-arabinofuranoside (**9**), 3,3'-di-O-methylellagic acid (**10**), and methyl 3,4,5-trihydroxybenzoate (**4**). The antioxidant activity of the isolated compounds were tested. Ellagic acid (**8**) exhibited potent 2,2-diphenyl-1-picrylhydrazyl (DPPH) scavenging activity with an SC₅₀ value of 4.05 μM.

Keywords: *Euscaphis japonica* var. *wupingensis*; Tannins; Chromones; DPPH scavenging activity

Contact information: a: College of Material Engineering, Fujian Agriculture and Forestry University, Fuzhou, Fujian 35002, China; b: Xiamen Botanical Garden, Xiamen, Fujian 361003, China; c: Furniture Products Quality Supervision and Testing Center of Putian, Xianyou, Fujian 351256, China; d: Fujian Provincial Key Laboratory of Innovative Drug Target Research, School of Pharmaceutical Sciences, Xiamen University, South Xiang-An Road, Xiamen 361102, China;

* Corresponding authors: cbangping@163.com; fjlingjg@126.com

INTRODUCTION

Euscaphis japonica var. *wupingensis* (Cai *et al.* 2018) is a variety in the bladdernut family Staphyleaceae. It is a deciduous shrub or small tree, glabrous, with leaves opposite, 5-7-pinnate, and it is generally found in the forests and mountains of Fujian, China. Its pericarp is yellow, and it can be distinguished easily from *E. japonica*.

E. japonica is a traditional Chinese medicinal material (Luo *et al.* 2012). Its roots, root bark, fruit, and flowers can be used as medicines because of a variety of effects (Li *et al.* 2016, 2018), and the pesticide effects are very different in the plant (Dong *et al.* 2004; Lee *et al.* 2009; Zhang *et al.* 2012). The fruits have the effect of treating headache, dizziness, and cold. The study of plant secondary metabolites is a crucial step in the scientific and rational use of plants (Yu *et al.* 2017). At present, 84 kinds of compounds have been isolated from *Euscaphis* Sieb. et Zucc., and the main active substances are triterpenoids, flavonoids, and phenolic acids (Tekeda *et al.* 2000; Hajime *et al.* 2009; Huang *et al.* 2014; Liang *et al.* 2018).

As a new variety, studies up to this point on *E. japonica* var. *wupingensis* have focused mainly on the morphological and biological characteristics, while its chemical properties have not been reported yet. In this paper, the chemical constituents from the fruits of this new variety were studied, which provides a theoretical basis for its rational utilization and functional development.

EXPERIMENTAL

Materials

Fruits of *E. japonica* var. *wupingensis* were collected from Huangtu hill (Wuping country, Fujian province, China) in October 2017. This site is located at 25°11'30" N, 116°02'37" E, altitude 400 to 520 m. The samples were oven-dried at 80 °C for 24 h and crushed before extraction.

The analytically pure solvents used in the extract and open-column separation process, including ethanol, methanol, and acetonitrile, were purchased from China National Pharmaceutical Corporation, Limited. The HPLC grade acetonitrile and methanol were obtained from Tedia (Fairfield, CT, USA). Ascorbic acid (Vc) and 2,2-diphenyl-1-picrylhydrazyl (DPPH) were purchased from J&K Scientific Co. Ltd. (Beijing, China).

General Equipment

The ¹H and ¹³C nuclear magnetic resonance (NMR) spectra were taken using TMS as the internal standard on a Bruker Avance III 600 FT NMR spectrometer (Bruker Corporation, Billerica, MA, USA). Chemical shifts were recorded as δ (ppm) values with dimethyl sulfoxide-*d*₆ (Sigma-Aldrich, St. Louis, MO, USA) as the solvent. Coupling constants (*J*) were recorded in Hz, and multiplicities were abbreviated as follows: s = singlet, d = doublet, t = triplet, br = broad, and m = multiplet.

The ESI-MS analysis was detected on a Thermo Q-Exactive Orbitrap Mass spectrometer (Thermo Fisher Scientific Corporation, Waltham, MA, USA), which was equipped with electrospray ionization source (ESI).

Column chromatography was performed on a Cosmosil 75 C18-OPN Column (75 μ m, Nakalai Tesque Co. Ltd., Kyoto, Japan). The preparative HPLC was performed with a Varian binary gradient LC system (Varian Inc. Corporate, Santa Clara, CA, USA) containing two solvent deliver modules (PrepStar 218), a photodiode array detector (ProStar 335), using an the preparative Cosmosil ODS column (250 mm \times 20.0 mm i.d., 5 μ m, Cosmosil, Nakalai Tesque Co. Ltd., Kyoto, Japan). A Sephadex LH-20 column (GE Healthcare, Sweden) was used for column chromatography with a glass column (120 cm \times 1.5 cm inner diameter); methanol was used as the eluent.

Extraction and Isolation

The fruits (100 g) were ultrasonically extracted with methanol for one hour (\times 3) under room temperature and evaporated until dry before weighing. The methanol extraction yielded 5.8 g of solid material having a brown color. For separation, 5.0 g of extract was chromatographed into a reversed-phase column, resulting in 5 fractions (abbreviated Fr. 1 – Fr. 5) with eluent of CH₃OH-water (20:80, 40:60, 60:40, 80:20, 100, v/v). Fr. 2 (2.5 g) was subjected to ODS column (25 g) and gradiently eluted with acetonitrile/H₂O (10% to 25%, 2L) to give 12 subfractions Fr. 2-1 – Fr. 2-12. Compound **1** was isolated from Fr. 2-6 (500 mg), through a preparative reversed-phase HPLC column,

and gradient-eluted with CH₃OH-water from 10:90 to 50:50 (v/v) at a flow of 8 mL/min. Fr. 3 (120 mg) was subjected to a Sephadex LH-20 column and eluted with methanol to afford compounds **6** and **7**. Further separation on Fr. 2-7 (800 mg), Fr. 2-9 (300 mg), and Fr. 2-12 (400 mg) by preparative HPLC resulted in the isolation of compounds **2**, **3**, and **5**, respectively. A preparative reversed-phase HPLC isocratic gradient eluted with CH₃OH-H₂O (0 – 60 min; 5% – 100 % CH₃OH, v/v) at a flow of 8 mL/min was used to isolate compounds **8**, **9**, and **10** from Fr. 4. The isolation scheme of these compounds is shown in Fig. 1.

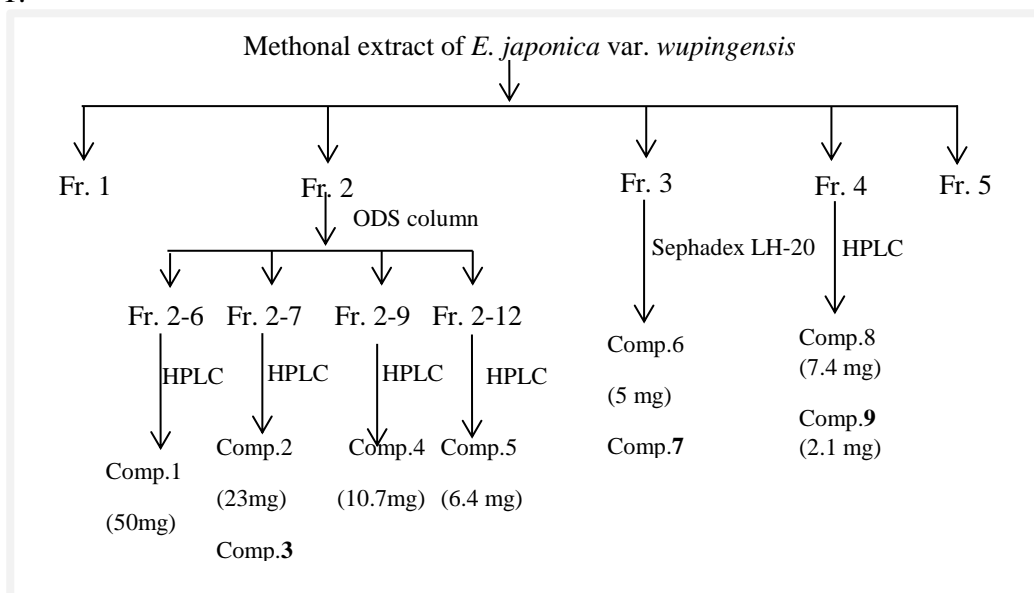


Fig. 1. Diagram of the extract and isolation of *E. japonica* var. *wupingensis*

DPPH Radical Scavenging Activity

The compound solutions of 50, 25, 12.5, 6.25, and 3.125 μ M were prepared with anhydrous ethanol, using vitamin C (Vc) as a positive control. DPPH radical scavenging activities of the isolated compounds were investigated according to the reported literature (Qiu *et al.* 2002). Briefly, a solution of the test compound in EtOH (0.1 mL), and 80 mM DPPH radical in EtOH (0.1 mL) was incubated at room temperature for 30 min. Reduction of the DPPH radical was measured at 517 nm. Measurements were performed in duplicate, and the concentration required for a 50% reduction (50% scavenging concentration, SC₅₀) of 40 mM DPPH radical was determined graphically, as shown in Table 2.

Isobiflorin, C₁₆H₁₈O₉ (**1**)

Compound **1** was obtained as colorless needles. ESI-MS: m/z 353 [M-H]⁻. The ¹H-NMR (DMSO-*d*₆, 600 MHz): δ_H 13.03 (1H, br.s, 5-OH), 6.25 (1H, s, H-6), 6.20 (1H, s, H-3), 4.63 (1H, d, $J = 9.5$ Hz, H-1'), 2.35 (3H, s, 2-CH₃); The ¹³C-NMR (DMSO-*d*₆, 151 MHz) spectral data are listed in Table 1.

Biflorin, C₁₆H₁₈O₉ (**2**)

Compound **2** was isolated as colorless needles. ESI-MS: m/z 353 [M-H]⁻. ¹H-NMR (DMSO-*d*₆, 600 MHz): δ_H 13.40 (1H, s, 5-OH), 6.39 (1H, s, H-8), 6.17 (1H, s, H-3), 4.56 (1H, d, $J = 9.7$ Hz, H-1'), 2.35 (3H, s, 2-CH₃); The ¹³C-NMR (DMSO-*d*₆, 151 MHz) spectral data are shown in Table 1.

Eugeniin, C₄₁H₃₀O₂₆ (3)

Compound **3** was obtained as a puce amorphous powder. ESI-MS: m/z 939 [M+H]⁺. ¹H-NMR (DMSO-*d*₆, 600 MHz): δ_H 6.88 (2H, s, galloyl-H), 6.87 (2H, s, galloyl-H), 6.81 (2H, s, galloyl-H), 6.76 (1H, s, HHDP-H), 6.28 (1H, s, HHDP-H), 5.68 (1H, t, $J = 9.9$ Hz, H-1), 5.53 (1H, t, $J = 9.7$ Hz, H-3), 5.37 (1H, br. d, $J = 8.5$ Hz, H-2), 5.12 (1H, dt, $J = 14.3, 7.0$ Hz, H-6), 4.55 (1H, br. dd, $J = 9.9, 7.2$ Hz, H-5), 3.83 (1H, br. d, $J = 13.0$ Hz, H-6). The ¹³C-NMR (DMSO-*d*₆, 151 MHz) spectral data were summarized as follows: δ 168.1 (C-7'''''), 168.1 (C-7'''''), 167.4 (C-7'), 165.9 (C-7''), 165.7 (C-7'''), 146.1 (C-3, 5'), 146.0 (C-3'', 5''), 145.9 (C-3''', 5'''), 145.7 (C-4''''', 6'''''), 145.7 (C-4''''', 6'''''), 139.4 (C-4'), 139.2 (C-4''), 139.2 (C-4'''), 135.9 (C-5'''''), 135.7 (C-5'''''), 124.8 (C-2'''''), 124.2 (C-2'''''), 119.2 (C-1'), 118.9 (C-1''), 118.8 (C-1'''), 115.8 (C-3'''''), 115.7 (C-3'''''), 109.3 (C-2', 2'', 2'''), 109.2 (C-6', 6'', 6'''), 106.2 (C-1'''''), 105.8 (C-1'''''), 98.8 (C-1), 73.5 (C-3), 72.7 (C-5), 72.1 (C-2), 70.6 (C-4), 63.0 (C-6).

Methyl 3,4,5-trihydroxybenzoate, C₈H₈O₅ (4)

Compound **4** was obtained as a colorless needles. ESI-MS: m/z 185 [M+H]⁺. ¹H-NMR (DMSO-*d*₆, 600 MHz): δ_H 6.94 (2H, s, H-2, 6), 3.74 (3H, s, CH₃). ¹³C-NMR (DMSO-*d*₆, 151 MHz): δ_C 166.8 (CO), 146.0 (C-3, 5), 138.9 (C-4), 119.7 (C-1), 108.9 (C-2, 6), 52.1 (CH₃).

2-Methyl-5,7-dihydroxy-chromone-7-O- β -D-glucopyranoside, C₁₆H₁₈O₉ (5)

Compound **5** was a light yellow powder. ESI-MS: m/z 355 [M+H]⁺, 353[M-H]⁻. ¹H-NMR (DMSO-*d*₆, 600 MHz): δ_H 12.83 (1H, br s, 5-OH), 6.66 (1H, d, $J = 2.0$ Hz, H-8), 6.42 (1H, d, $J = 2.2$ Hz, H-6), 6.26 (1H, s, H-3), 5.04 (1H, d, $J = 7.5$ Hz, H-1'), 2.39 (3H, s, 2-CH₃); The ¹³C-NMR (DMSO-*d*₆, 151 MHz) spectral data are shown in Table 1.

5,7-Dihydroxy-2-methyl-4H-chromen-4-one, C₁₀H₈O₄ (6)

Compound **6** was a light yellow crystal. ESI-MS: m/z 193[M+H]⁺. ¹H-NMR (DMSO-*d*₆, 600 MHz): δ_H 6.33 (1H, d, $J = 2.0$ Hz, H-3), 6.17 (2H, s, H-6, 8), 2.35 (3H, s, 2-CH₃). The ¹³C-NMR (DMSO-*d*₆, 151 MHz) spectral data are shown in Table 1.

Quercetin-3-O- β -D-arabinoside, C₂₀H₁₈O₁₁ (7)

Compound **7** was a yellow powder. ESI-MS: m/z 435 [M+H]⁺. ¹H-NMR (DMSO-*d*₆, 600 MHz): δ_H 12.63 (1H, s, 5-OH), 7.65 (1H, dd, $J = 2.0, 8.4$ Hz, H-6'), 7.53 (1H, d, $J = 2.2$ Hz, H-2'), 6.85 (1H, d, $J = 8.4$ Hz, H-5'), 6.39 (1H, d, $J = 2.0$ Hz, H-8), 6.21 (1H, d, $J = 2.0$ Hz, H-6), 5.26 (1H, d, $J = 5.3$ Hz, H-1''); ¹³C-NMR (DMSO-*d*₆, 151 MHz): δ_C 177.9 (C-4), 164.9 (C-7), 161.6 (C-5), 156.7 (C-2), 156.7 (C-9), 149.1 (C-4'), 145.5 (C-3', 5'), 134.1 (C-3), 122.4 (C-2', 6'), 121.3 (C-1'), 116.2 (C-3', 5'), 115.9 (C-2', 6'), 104.3 (C-10), 101.9 (C-1''), 99.2 (C-6), 94.0 (C-8), 72.1 (C-3''), 71.2 (C-2''), 66.6 (C-4''), 64.8 (C-5'').

Ellagic acid, C₁₄H₆O₈ (8)

Compound **8** was a grey powder. ESI-MS: m/z 301 [M+H]⁺. ¹H-NMR (DMSO-*d*₆, 600 MHz): δ_H 7.45 (2H, s, H-5,5'); ¹³C-NMR (DMSO-*d*₆, 151 MHz): δ_C 159.9 (C-7,7'), 148.9 (C-4,4'), 136.8 (C-3,3'), 130.1 (C-2,2'), 113.1 (C-6,6'), 110.1 (C-5,5'), 106.7 (C-1,1').

3,3'-Di-O-methylellagic acid 4-(5''-acetyl)- α -L-arabinofuranoside, $C_{23}H_{20}O_{13}$ (**9**)

Compound **9** was obtained as white amorphous powder. ESI-MS: m/z 503 [M-H]⁻. ¹H-NMR (DMSO-*d*₆, 600 MHz): δ_H 7.73 (1H, s, H-5), 7.52 (1H, s, H-5'), 5.68 (1H, d, H-1''), 4.08 (3H, s, 3-OCH₃), 4.05 (3H, s, 3'-OCH₃), 3.87 (1H, dd, J=6 and 4 Hz, H-3''), 2.04 (3H, s, 6''-CH₃); ¹³C-NMR (DMSO-*d*₆, 151 MHz): δ_C 170.7 (C-6''), 159.1 (C-7), 142.5 (C-3, 2), 142.2 (C-3', 2'), 113.3 (C-5, 5'), 112.4 (C-6, 6'), 108.0 (C-1''), 83.1 (C-4''), 82.2 (C-2''), 77.4 (C-3''), 64.1 (C-5''), 62.0 (3-OCH₃), 61.3 (3'-OCH₃), 21.1 (6''-CH₃).

3,3'-Di-O-methylellagic acid, $C_{16}H_{10}O_8$ (**10**)

Compound **10** was obtained as yellow powder. ESI-MS: m/z 329 [M-H]⁻. ¹H-NMR (DMSO-*d*₆, 600 MHz): δ_H 7.51 (2H, s, H-5, 5'), 4.04 (6H, s, 3, 3'-OCH₃); ¹³C-NMR (DMSO-*d*₆, 151 MHz): δ_C 159.0 (C-7,7'), 152.9 (C-4,4'), 140.8 (C-3,3'), 141.7 (C-2,2'), 112.6 (C-6, 6'), 112.0 (C-5, 5', 1, 1'), 61.4 (3,3'-OCH₃).

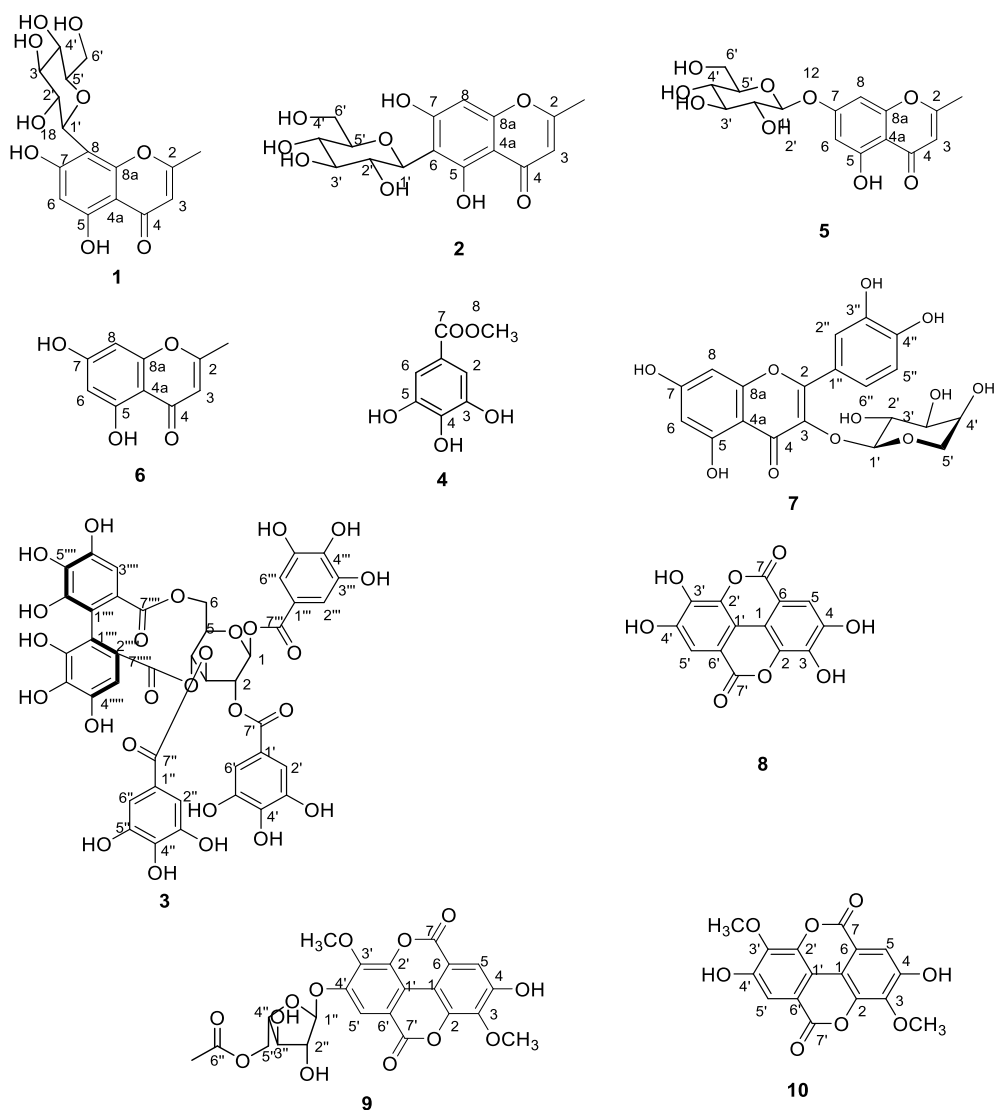


Fig. 2. Chromones and tannins from the fruit of *Euscaphis japonica* var. *wupingensis*

Table 1. ^{13}C -NMR (DMSO- d_6 , 151 MHz) data of compounds **1**, **2**, **5**, and **6**

Number	1	2	5	6
2	167.9	167.9	161.6	168.1
3	108.4	108.1	108.8	108.4
4	182.4	182.6	182.5	182.2
5	161.2	160.9	157.9	164.6
6	109.2	98.7	100.3	99.2
7	163.9	163.2	168.9	162.0
8	93.9	104.95	95.0	94.2
9	157.2	156.87	163.4	158.3
10	103.5	104.1	105.5	103.9
2-CH ₃	20.3	20.3	20.5	20.4
1'	73.4	73.58	99.9	
2'	71.1	71.4	73.5	
3'	79.4	79	77.6	
4'	70.6	70.9	70.0	
5'	82.1	81.8	76.8	
6'	62.0	61.9	61.1	

RESULTS AND DISCUSSION

Phytochemistry of the Fruit of *E. japonica* var. *wupingensis*

Isolation of the fruit of *E. japonica* var. *wupingensis* yielded five chromones, four tannins, and methyl 3,4,5-trihydroxybenzoate. Their chemical structures were determined by their ESI-MS, ^1H -NMR, and ^{13}C -NMR spectra. Two chromones, 2-methyl-5,7-dihydroxy-chromone-7-*O*- β -D-glucopyranoside (**5**) and quercetin-3-*O*- β -D-arabinoside (**7**), and two tannin, eugenin (**3**) and 3,3'-di-*O*-methylellagic acid 4-(5''-acetyl)- α -L-arabinofuranoside (**9**), were isolated from the family of Staphyleaceae for the first time. The structures of the isolated compounds are shown in Fig. 2.

Compound **3** was isolated as a puce amorphous powder. The dark color, high polarity, as well as the high molecular weight, induced by the ESI-MS quasio-molecular ion peak at m/z 939 $[\text{M}+\text{H}]^+$, indicated that compound **3** is a tannin derivate. By comparing its ^{13}C -NMR data with those reported in literature (Lian *et al.* 2017), the structure of **3** was determined as eugenin.

Compound **5** was obtained as a light yellow powder. In low field of ^1H -NMR (DMSO- d_6 , 600 MHz) of **5**, the signal of δ_{H} 12.83 (1H, br.s) belongs to the C-5 hydroxyl, forming intramolecular hydrogen bond with C=O. A pair of AM spin system aromatic proton signals were found at δ_{H} 6.66 (1H, d, $J = 2.0$ Hz) and 6.42 (1H, d, $J = 2.2$ Hz), belonging to the H-8 and H-6 of the chromone's ring. The signal at δ_{H} 6.26 (1H, s) was assigned to H-3. The high field of ^1H -NMR shown the presence of methyl at δ_{H} 2.39 (3H, s, 2-CH₃). The ^{13}C -NMR revealed an oxy-glycosylated glucose, based on the signals at δ_{C} 95.0 (C-1'), 77.6 (C-3'), 76.8 (C-5'), 73.5 (C-2'), 70.0 (C-4') and 61.1 (C-6'). According to the presence of 5-OH and absence of 7-OH ^1H -NMR signal, the glucosyl was considered to be attached to C-7. Thus, compound **5** was 2-methyl-5,7-dihydroxychromanone glucoside. The above data is consistent with the reported NMR data (Yang *et al.* 2006). Therefore, the compound was identified as 5,7-dihydroxy-2-methylchromanone-7-*O*- β -D-glucoside.

Compound **7** was isolated as yellow powder. Both the hydrochloric acid-magnesium powder (HCl-Mg) reaction and the Molish reaction were positive, indicating that the **7** is a flavonoid glycoside. In the $^1\text{H-NMR}$ spectrum, the signal of 5-OH was shown at δ_{H} 12.63 (1H, s, 5-OH). The AMX spin system signals at δ_{H} 7.65 (1H, d, $J = 2.0$ Hz), 7.64 (1H, d, $J = 2.2$ Hz), 7.53 (1H, d, $J = 2.2$ Hz) suggests the 3',4'-disubstituted B flavonoid ring; The meta-coupled benzene ring signals at δ_{H} 6.39 (1H, d, $J = 2.0$ Hz) and 6.21 (1H, d, $J = 2.0$ Hz) belonged to H-8 and H-6 in the A ring, respectively. The signal at δ_{H} 5.26 (1H, d, $J = 5.3$ Hz, H-1") was attributed to a terminal proton of the sugar. By comparing the $^{13}\text{C-NMR}$ data with those in the literature (Jiang *et al.* 2009), compound **7** was identified as quercetin 3-O- α -L-arabinopyranoside.

Compound **9** was obtained as white amorphous powder. The $^1\text{H-NMR}$ spectrum showed 9 proton signals, including two benzene ring signals at δ_{H} 7.73, 7.52; two methoxy signals at δ_{H} 4.08, 4.05; an acetyl signal at δ_{H} 2.04, and a set of glycoside proton signal ranging from δ_{H} 5.68 to 3.87. The $^{13}\text{C-NMR}$ shows the unsaturated lactone signal at δ 159.1; an arabinofuranose signal δ_{C} 108.1, 83.1, 82.2, 77.4, and 64.1. $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ indicate that the compound is a steroidal compound linked to a glycoside; the $^{13}\text{C-NMR}$ signals at δ_{C} 62.0 and δ 61.3 suggest methoxy groups. The NMR data were similar to the reported results (Tanaka *et al.* 2001) (Jules *et al.* 2007). Therefore, compound **9** is identified as 3,3'-di-O-methylellagic acid 4-(5"-acetyl)- α -L-arabinofuranoside.

By comparing NMR data with those reported in literature, other known compounds were identified as: isobiflorin (**1**) (Takashi *et al.* 1993) (Lee *et al.* 2016), biflorin (**2**) (Li *et al.* 2009), methyl 3,4,5-trihydroxybenzoate (**4**) (Yang *et al.* 1998), 5, 7-dihydroxy-2-methyl-4H-chromen-4-one (**6**) (Cao *et al.* 2005), ellagic acid (**8**) (Xiao *et al.* 2017), and 3,3'-di-O-methylellagic acid (**10**) (Nono, *et al.* 2014).

Antioxidant Activity

Table 2 shows that the chromones and quinones in the extract of *E. japonica* var. *wupingensis* have a good scavenging effect on DPPH free radicals. Higher concentrations resulted in a stronger ability to scavenge free radicals. Ellagic acid (**8**, SC_{50} of 4.05 μM) has the strongest effect on scavenging DPPH freeness, which is twice as high as that of vitamin C (SC_{50} of 9.85 μM).

Table 2. Antioxidant Activity of Compounds Isolated from *E. japonica* var. *wupingensis*

Compounds	SC_{50} (μM)
Isobiflorin (1)	14.20
biflorin (2)	22.06
2-methyl-5,7-dihydroxy-chromone-7-O- β -D-glucopyranoside (5)	19.48
Quercetin-3-O- β -D-arabinoside (7)	14.88
Ellagic acid (8)	4.05
3,3'-di-O-methylellagic acid (10)	29.57
Vc	9.85

CONCLUSIONS

- E. japonica* var. *wupingensis* was extracted with methanol, and the extract components were isolated using column chromatography. Five chromones, four tannins, and methyl 3,4,5-trihydroxybenzoate were identified as isobiflorin (**1**), biflorin (**2**), 2-methyl-5,7-

dihydroxy-chromone-7-O- β -D-glucopyranoside (5), 5,7-dihydroxy-2-methyl-4H-chromen-4-one (6), quercetin-3-O-D-arabinoside (7), eugenin (3), ellagic acid (8), 3,3'-di-O-methylellagic acid 4-(5"-acetyl)- α -L-arabinofuranoside (9), and 3,3'-di-O-methylellagic acid (10).

2. For the first time, the chemical composition isolated from the *Euscaphis japonica* var. *wupingensis* is reported. Two chromones (5 and 7) and two tannins (3 and 9) were first isolated from the genus of *Euscaphis*.
3. Ellagic acid (8) has stronger antioxidant activity than vitamin C, and the specific efficacy needs to be further studied.

ACKNOWLEDGMENTS

The authors are grateful for the funding from Fujian Provincial Department of Finance Research Fund "Research and Extension of Key Technique on Characteristic and Identification of Import Wood" (No. K8115004A) and Fujian Provincial Science and Technology Bureau Fund "Germplasm collection and space mutation breeding of medicinal and ornamental plants of *Euscaphis japonica*" (No. 3502Z20144073).

REFERENCES CITED

- Cai, B. P., Zhang X. Y., Guo, H. Z., Liu, S. Z., and Huang, K. F. (2018). "*Euscaphis japonica* var. *wupingensis*, a new variety with yellow pericarp from Fujian, China," *Phytotaxa* 334(1), 55-59. DOI: 10.11646/phytotaxa.334.1.8
- Cao, P., Pu, X. F., Peng, S. L., Zhang, X. R., and Ding, L. S. (2005). "Chemical constituents from *Cimicifuga foetida*," *Journal Asian Natural Products Research* 7, 145-149. DOI: 10.1080/1028602042000204081
- Dong, M., Zhang, Q. X., and Guang, T. M. (2004). "Study on the anti-inflammatory and chemical structure of euscapholide," *Natural Product Research and Development* 16(4), 290-293.
- Hajime, M., Yosuke, M., Takashi, T., and Isao, K. (2009). "Euscaphinin, a new ellagitannin dimer from *Euscaphis japonica* (THUNB.) KANITZ," *Chemical & Pharmaceutical Bulletin* 57(4), 421-423. DOI: 10.1248/cpb.57.421
- Huang, Y., Xiang, D. B., Hu, Q. M., Tan, Y., Men, Y. C., and Pei, G. (2014). "Phenolic acids from fruits of *Euscaphis fukienensis*," *Chinese Traditional and Herbal Drugs* 45, 2611-2613. DOI: 10.7501/j.issn.0253-2670.2014.18.007
- Jiang, J. L., and Shi, R. (2009). "A new acylated quercetin glycoside from the leaves of *Stevia rebaudiana* Bertoni," *Nature Product Research* 23, 1378-1383. DOI: 10.1080/14786410802447294
- Jules, D. J., Eliane, A. M., Leon, A. T., David, L., and Raffaele, T. (2007). "Identification of ellagic acid derivatives from stem bark of *Syzygium guineense* (Myrtaceae)," *Natural Product Communications* 2, 1-6.
- Lee, M. K., Lee, K. Y., Jeon, H. Y., Sung, S. H., and Kim, Y. C. (2009). "Antifibrotic activity of triterpenoids from the aerial parts of *Euscaphis japonica* on hepatic stellate cells," *Journal of Enzyme Inhibition and Medicinal Chemistry* 24, 1276-1279. DOI: 10.3109/14756360902829709

- Lee, H. H., Shin, J. S., Lee, W. S., Yu, B., Jang, D. S., and Lee, K. T. (2016). "Biflorin, isolated from the flower buds of *Syzygium aromaticum* L., suppresses LPS-induced inflammatory mediators via STAT1 inactivation in macrophages and protects mice from endotoxin shock," *Journal of Natural Products* 79, 711-720. DOI: 10.1021/acs.jnatprod.5b00609
- Liang, W. X., Ni, L., Zhou, X. X., Huang, W., and Zhou, S. Q. (2018). "Research progress on chemical constituents of *Euscaphis* and their pharmacological effects," *Chinese Traditional and Herbal Drugs* 49(5), 1220-1226. DOI: 10.7501/j.issn.0253-2670.2018.05.034
- Lian, Z. H., Liu, W., Zheng, J., Xu, R., Du, H. G., and Liu, A. (2017). "Isolation, structural characterization and neuraminidase inhibitory activities of polyphenolic constituents from *Flos caryophylli*," *Phytochemistry Letters* 19, 160-167. DOI: 10.1016/j.phytol.2016.12.031
- Li, J., Jiang, H., and Shi R. (2009). "A new acylated quercetin glycoside from the leaves of *Stevia rebaudiana* Bertoni," *Natural Product Research* 23 (15), 1378-1383. DOI: 10.1080/14786410802447294
- Li, Q., Li, Q. Y., Lin, J. G.; Li, J. Q., Wu, H., Liu, J., Wang, X. X. (2016). "Decay resistance effects of *Pinus massoniana* treated with different preservatives based on pyrolysis and thermodynamics," *Wood Sciences Technology* 50, 105-116. DOI: 10.1007/s00226-015-0780-2
- Li, Q., Xu, L., Wu, H., Liu, J., Lin, J., Guan, X. (2018). "Differential proteome analysis of the extracts from the xylem of *Cinnamomum camphora* inhibiting *Coriolus versicolor*," *Holzforschung* 72, 459-466. DOI:10.1515/hf-2017-0148
- Luo, H. Y., Yao, M., Shen, W. X., Jia, X., Li, W. J., Yang, Y., Gong, J., and Ni, S. F. (2012). "Overview of pharmacological research on *Euscaphis*," *Journal of Anhui Agricultural Sciences* 40(15), 8462-8463. DOI: 5017-6611(2012)15-08462-02
- Nono, R. N., Barboni, L., Teponno, R. B., Quassinti, L., Bramucci, M., Vitali, L. A., Petrelli, D., Lupidi, G., and Taponjou, A. L. (2014). "Antimicrobial, antioxidant, anti-inflammatory activities and phytoconstituents of extracts from the roots of *Dissotis thollonii* Cogn. (Melastomataceae)," *South African Journal of Botany* 93, 19-26. DOI: 10.1016/j.sajb.2014.03.009
- Qiu, Y. K., Chen, Y. J., and Pei, Y. P. (2012). "Constituents with radical scavenging effect from *Opuntia dillenii*: Structures of new α -pyrones and flavonol glycoside," *Chemical & Pharmaceutical Bulletin* 50(11), 1507-1510. DOI: 10.1248/cpb.50.1507.
- Tanaka, T., Orii, Y., Nonaka, G. I., and Nishioka, I. (1993). "Tannins and related compounds. (CXXIII. 1a) Chromone, acetophenone and phenylpropanoid glucosides and their galloy and/or hexahydroxydiphenoyl esters from the leaves of *Syzygium aromaticum* Merr. et Perry," *Chemical & Pharmaceutical Bulletin* 41(7), 1232-1237.
- Tanaka, N., Tanaka, T., Fujioka, T., Fujii, H., Mihashi, K., Shimomura, K., and Ishimaru, K. (2001). "An ellagic compound and iridoids from *Cornus capitata* root cultures," *Phytochemistry* 57 (8), 1287-1291. DOI: 10.1016/S0031-9422(01)00179-0
- Tekeda, Y., Okada, Y., Masuda, T., Hirata, E., Shinzato, A., Takushi, A., Yu, Q., and Otsuka, H. (2000). "New megastigmane and tetraketide from the leaves of *Euscaphis japonica*," *Chemical & Pharmaceutical Bulletin* 48(5), 752-754. DOI: 10.1002/chin.200042208
- Xiao, T., Guo, Z. H., Bi, X. L., and Zhao, Y. Q. (2017). "Polyphenolic profile as well as anti-oxidant and anti-diabetes effects of extracts from freeze-dried black raspberries," *Journal of Functional Foods* 31, 179-187. DOI: 10.1016/j.jff.2017.01.038

- Yang, J., Pan, Q., Wei, D. F., and Wang, Y. (2006). "Studies on chemical constituents of *Berchemia polyphylla* var. *leioclada*," *Chinese Pharmaceutical Journal* 41 (4), 255-257.
- Yang, X., Gu, Z. M., Ma, C. M., Masao, H., and Tsuneo, N. (1998). "A new indole derivative isolated from the root of tuber fleece flower (*Polygonum multiflorum*)," *Chinese Traditional and Herbal Drugs* 29(1), 5-11.
- Yu, J. Q., Song, X. Y., Wang, D. J., Wang, X. J., and Wang, X. (2017). "Five new chromone glycosides from *Scindapsus officinalis* (Roxb.) Schott," *Fitoterapia* 122, 101-106. DOI: 10.1016/j.fitote.2017.09.002
- Zhang, L. J., Cheng, J. J., Liao, C. C., Cheng, H. L., Huang, H. T., Kuo, L. M., and Kuo, Y. H. (2012). "Triterpene acids from *Euscaphis japonica* and assessment of their cytotoxic and anti-NO activities," *Planta Medica* 78, 1584-1590. DOI: 10.1055/s-0032-1315040

Article submitted: January 25, 2019; Peer review completed: March 23, 2019; Revised version received: April 2, 2019; Accepted: May 16, 2019; Published: May 22, 2019.
DOI: 10.15376/biores.14.3.5355-5364