

## Potential Use of Green Alga *Ulva* sp. for Papermaking

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The large amount of cellulose found in *Ulva* sp. and its low percentage of lignin-like compounds make it an interesting raw material for partially substituting wood pulp to produce pulp and paper. This work shows the suitability of mild chemical treatments for papermaking using residual biomass from this green seaweed, harvested on the beaches, in order to give it added value. A chemical characterization was used to determine ethanol-benzene, hot water, and 1% soda extractives, ash content, holocellulose,  $\alpha$ -cellulose, and acid-insoluble material. Cellulose extraction was performed with low proportions of soda and hydrogen peroxide, and it was subjected to a refining step. A design of experiments was used to explain the influence of soda (6%, 8%, and 10%) and hydrogen peroxide (2%, 4%, and 6%) based on oven-dry weight, plus refining (1000 PFI revolutions, 3000 PFI revolutions, and 5000 PFI revolutions). The results showed that to attain good paper strength, it is advisable to operate at maximum alkali charge, minimum peroxide concentration, and refine to a high degree.

**Keywords:** *Ulva* sp.; Green algae; Pulping; Refining; Hydrogen peroxide; Sodium hydroxide

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## INTRODUCTION

Currently, there is increased interest in producing paper from seaweed. To a large extent, the potential use of seaweed is motivated by the clear advantages that they present with respect to terrestrial biomass, such as their very low or null lignin content and rapid growth. The cellulose contents are close to those of conventional fiber sources, which gives them great importance for their use in obtaining paper (Knoshaug *et al.* 2013). Numerous benefits have been reported from seaweed's use in papermaking, as it avoids the dependence on tree monocultures and the variability of the price of wood sources, which is expected to increase over time. Furthermore, it involves low consumption of chemical reagents due to its easy treatment, valorization of coastal waste associated with environmental and economic problems in tourist areas, and higher productivity of biomass than non-fertilized crops (Rajkumar *et al.* 2014; Baweja *et al.* 2016). It has certain drawbacks such as the difficulty in removing salts, inorganic compounds, and other residues. This can make the production process expensive. In some cases, the sheets of paper made from seaweed have poor mechanical properties (Ververis *et al.* 2007; Khiari *et al.* 2010).

A wide variety of algae have been used to obtain paper with different pulping treatments with a low environmental impact (Knoshaug *et al.* 2013). You and Park (2004) patented a method for producing paper from *Rhodophytae* by immersing the algae in an alcohol-based extraction solvent followed by boiling. The extraction can be carried out without chemicals at 120 °C to 140 °C for *Gelidialian* sp. (Seo *et al.* 2010), or with 2%

NaOH at 100 °C in the case of *Gracilaria verrucosa* (Laksitoesmi *et al.* 2010). Pulp and paper have been obtained using green algae, requiring low doses of soda and temperatures around 100 °C for *Chaetomorpha* sp. (Kiran *et al.* 1980) or *Rhizoclonium* sp. (Chao *et al.* 2000). Nicolucci *et al.* (1995, 1996) explained a simple physical treatment that reduces the particle size followed by a chemical treatment at 70 °C with soda and hydrogen peroxide in low concentrations to obtain paper from a mixture of green and red algae (*Ulva rigida*, *Ulva lactuca*, *Enteromorpha intestinalis*, and *Gracilaria confervoides*). However, to the best of the authors' knowledge, the green algae *Ulva* has not been used alone for papermaking by environmentally sustainable methods.

Green algae are promising raw materials for papermaking as they are known to contain more cellulose than brown or red algae (Baweja *et al.* 2016; Jmel *et al.* 2016). Moreover, they are a good source of dietary fiber and other important nutrients such as amino acids and polyunsaturated fatty acids, comparable to fruits and vegetables. The possibility of using *Ulva pertusa* as feedstock for biofuel (Choi *et al.* 2013; Kim and Ha 2015; Michalak 2018) and for the bioproduction of D- glucaric acid and succinic acid has been established (Lee *et al.* 2014). *U. prolifera* has been investigated for bioethanol production using a pretreatment with hydrogen peroxide and enzymatic hydrolysis (Li *et al.* 2016), and *Ulva lactuca* has been valorized by the extraction of insoluble fiber and soluble dietary fiber (Yaich *et al.* 2011). Cellulose extracted from green seaweed *Ulva fasciata* was processed to synthesize biodegradable carboxymethyl cellulose films with applications in the cosmetic, food, textile, medical, agricultural, and pharmaceutical industries (Lakshmi *et al.* 2016). Likewise, cellulose from *Cladophora* sp. algae has been proposed to prepare a conducting polypyrrole-based composite material (Mihiranyan *et al.* 2008; Olsson *et al.* 2012) for use in ion exchange membranes or sensors.

In this work, the main objective was to obtain a product that resembles the characteristics of conventional paper using the green algae *Ulva* sp. as a source of cellulose fiber, together with conventional softwood kraft fibers (50%). This approach has been chosen by other authors (Seo *et al.* 2010), and it improves inter-fiber bonding because the size and shape of algal fibers might be too homogeneous. A mild chemical treatment with low-polluting reagents (soda and hydrogen peroxide) was optimized by following a central composite experimental design. A chemical characterization of *Ulva* sp. waste was conducted to determine its potential for pulp and paper production. Finally, mechanical and optical tests of the paper sheets were performed, and the effects of the ratio of soda to dry material, the ratio of hydrogen peroxide to dry material, and refining were studied.

## EXPERIMENTAL

### Materials

*Ulva* sp. from tidal waste was harvested by hand at “Punta Entinas-Sabinar” (Almería, Spain) in September 2017. Samples were washed with freshwater, screened to remove sand and other impurities, and dried at 40 °C for 3 days. Diethylene-triaminepentaacetic acid (DTPA) and anthraquinone were supplied by Sigma-Aldrich (Madrid, Spain). Anhydrous MgSO<sub>4</sub>, NaOH pellets, and hydrogen peroxide were purchased from PanReac (Barcelona, Spain).

### Cellulose Extraction

*Ulva* sp. was cooked in a 15 L batch cylindrical reactor with NaOH (6%, 8%, and 10%), MgSO<sub>4</sub> (0.2%), and anthraquinone (1%) at 90 °C for 60 min, adding water to achieve

a liquid-to-solid ratio of 9. It should be noted that the percentages express the weight ratio of reagent to oven-dry raw material, not to the solution.

The resulting pulps were treated with hydrogen peroxide (2%, 4%, and 6%, based on oven-dry pulp weight) and 0.5% DTPA (Abdel-Halim and Al-Deyab 2013). Each sample was placed in a polyethylene bag and immersed in a thermostatic bath at 60 °C for 60 min. Finally, samples were dried at 40 °C for one week.

## Refining

Cellulose pulps were beaten at 10% consistency in a Metrotec PFI mill (Lezo, Spain). The numbers of revolutions applied were 1000, 3000, and 5000. The electric power of the engine was 0.37 kW as the voltage applied was 220 V. The degree of refining, expressed as freeness, was measured using a Canadian Standard Freeness (CSF) tester (West Berlin, NJ, USA), following TAPPI T227 om-17 (2017).

## Chemical Characterization of the Samples

The pulp obtained was characterized chemically in accordance with the common TAPPI test methods for raw materials and/or pulps (TAPPI 2019), beginning with T264 cm-07 (2007) for the preparation of samples for chemical analysis. The standard T203 cm-09 (2009) was followed to estimate the content of alpha-cellulose, understood as the fraction that is resistant to consecutive treatments with 17.5% and 9.45% NaOH solutions, after a chlorite delignification was performed to measure the holocellulose content (Ahlgren and Goring 1971). According to T222 om-15 (2015), the determination of acid-insoluble Klason lignin was carried out with H<sub>2</sub>SO<sub>4</sub> 24N. Solid-liquid extractions followed T204 cm-17 (2017) for ethanol-benzene extractives and T207 cm-08 (2008) for hot water solubility, while the ash content was determined by means of a muffle furnace in accordance with T211 sp-11 (2011).

## Sheet Forming and Testing

Ten isotropic handsheets were made from each of the *Ulva* sp. samples, always mixed with an unbleached pine kraft pulp (PKP) in a 1:1 weight ratio. The PKP, reportedly obtained from the wood of *Pinus pinaster* Ait., was of industrial origin. In any case, a laboratory sheet former conforming to the ISO standard 5269-1 (2005) was used. Agitation was done by hand, with a standard stirrer. Couch weights and standard plates were used to collect the handsheets. Sheets were left at 23 °C and 50% RH, while pressed by drying rings, for 24 h. The basis weight was 60 g/m<sup>2</sup>.

The tensile test for the breaking length and stretch, the burst test, and the tear test were performed by means of appropriate testing machines from Hounsfield, Metrotec and Messmer, respectively, and in accordance to the ISO standards 1924 (1992), 1974 (2012), and 2758 (2014). Brightness was determined by means of a spectrophotometer from Lorentzen & Wettre (Kista, Sweden), following ISO 2470 (2009).

## Experimental Design

In order to evaluate the influence of three independent variables (alkali dose, hydrogen peroxide concentration and refining) on the physical properties of sheets, a central composite design with response surface methodology was chosen as the approach (Abd Hamid *et al.* 2014). This experimental design consisted of 15 tests that were used to estimate the quadratic terms of a polynomial model (Table 1). The values of the independent variables were normalized by using Eq. 1 in order to facilitate the direct

comparison of coefficients and expose the individual effects of the independent variables on each dependent variable,

$$X_n = 2 \frac{X - \bar{X}}{X_{max} - X_{min}} \quad (1)$$

where  $X_n$  is the normalized value of soda ( $S$ ), hydrogen peroxide ( $P$ ), or refining degree ( $R$ );  $X$  is the absolute experimental value of the variable concerned;  $\bar{X}$  is the mean of the extreme values; and  $X_{max}$  and  $X_{min}$  are its maximum and minimum value, respectively. The variables values for each test are shown in Table 1.

**Table 1.** Experimental Design: Sets of Conditions and Normalized Values

Assay	NaOH (%)	H <sub>2</sub> O <sub>2</sub> (%)	PFI revolutions	X <sub>S</sub>	X <sub>P</sub>	X <sub>R</sub>
1	8	4	3000	0	0	0
2	10	6	5000	1	1	1
3	6	6	5000	-1	1	1
4	10	6	1000	1	1	-1
5	6	4	1000	-1	1	-1
6	10	2	5000	1	-1	1
7	6	2	5000	-1	-1	1
8	10	2	1000	1	-1	-1
9	6	2	1000	-1	-1	-1
10	8	6	3000	0	1	0
11	8	2	3000	0	-1	0
12	8	4	5000	0	0	1
13	8	4	1000	0	0	-1
14	10	4	3000	1	0	0
15	6	4	3000	-1	0	0

Experimental data were fitted to a second-order polynomial model (Eq. 2) which relates each dependent variable (tear index, breaking length, stretch, burst index, and brightness) with the operational variables.

$$Z = a_0 + \sum_{i=1}^n b_i X_{ni} + \sum_{i=1}^n c_i X_{ni}^2 + \sum_{i=1; j=1}^n d_{ij} X_{ni} X_{nj} (i < j) \quad (2)$$

In Eq. 2,  $Z$  and  $X_{ni}$  denote dependent and normalized independent variables, respectively, and  $a_0$ ,  $b_i$ ,  $c_i$ , and  $d_{ij}$  are constants to be estimated from the experimental data. For three independent variables, the model is presented in Eq. 3,

$$Z = a_0 + b_1 X_S + b_2 X_P + b_3 X_R + c_1 X_S^2 + c_2 X_P^2 + c_3 X_R^2 + d_{12} X_S X_P + d_{13} X_S X_R + d_{23} X_P X_R \quad (3)$$

where  $Z$  stands for each of the dependent variables. The software SigmaPlot® (Version 12.0) was used to formulate the models and analyze the data.

## RESULTS AND DISCUSSION

### Chemical Composition of *Ulva* sp. Waste

The results of the chemical characterization of *Ulva* sp. dead biomass, including ethanol-benzene extractives (EBE), hot water and 1% soda solubilities (HWS and 1% SS), ash, holocellulose (HOL),  $\alpha$ -cellulose ( $\alpha$ -CEL), and acid insoluble material (KLAS), are presented in Table 2. For comparison purposes, chemical composition data from a different green alga, from another non-wood material, and from a conventional raw material for papermaking are shown. The EBE content is very low in typical raw materials for papermaking such as *Eucalyptus globulus* or *Pine pinaster* (1.2% and 2.6%, respectively; Jiménez *et al.* 2007), and the values found for this macroalga ( $3.8\% \pm 1.1\%$ ) were close to those in wood. In contrast, HWS and 1% SS were much higher than those of wood and woody stems (Jiménez *et al.* 2007), but similar to those of certain grasses (Sharma *et al.* 2018). This was probably due to the starch fraction (Mukherjee and Keshri 2019). However, the high ash content in *Ulva* sp. ( $19.8\% \pm 3.1\%$ ) is especially remarkable. This result is comparable with previous studies on *Ulva lactuca* (19.6%; Yaich *et al.* 2011), *Ulva pertusa* Kjellman (22.5%; Lee *et al.* 2014) or other green algae such as *Rhizoclonium* sp. (15.9%; Chao *et al.* 1999). It is clearly low in comparison with brown algae (30 to 40%; Rupérez 2002). In general, macroalgae are characterized by high ash content and a very large percentage of mineral salts that is much higher than for vascular plants. When algae are collected, they contain sand and carbonated deposits. These impurities are largely eliminated during cleaning, but the residual contaminants contribute to the high percentage of ash. The ash contents in seaweeds are variable, and many factors such as the species or location, collection date, salinity, *etc.*, determine its value (Dawes *et al.* 1987).

Cellulose, the main component of the cell wall, is more abundant in green algae than in red or brown algae (Kim and Ha 2015; Jmel *et al.* 2016). In this study, holocellulose was the most abundant fraction in *Ulva* (Table 2). Yaich *et al.* (2015) observed that hemicellulose was the main fraction (32.5%) of *U. lactuca*, followed by  $\alpha$ -cellulose (16.6%). The  $\alpha$ -CEL content of *Ulva* sp. dead biomass was higher than those obtained for *U. prolifera*, 19.4%  $\alpha$ -cellulose, and 14.4% hemicellulose (Li *et al.* 2016), or *U. pertusa* Kjellmann, with 6.7% cellulose and 16.8% hemicellulose (Choi *et al.* 2013).

**Table 2.** Chemical Characterization of *Ulva* sp. and Comparison with Other Materials

Parameter	<i>Ulva</i> sp.		<i>Rhizoclonium</i> sp. (alga)	<i>C. winterianus</i> (grass)
	Average	SD		
HWS (%)	33.4	1.67	34.6	21.5
1% SS (%)	29.8	0.47	36.3	28.2
EBE (%)	3.8	0.47	9.43	6.31
ASH (%)	19.8	2.76	15.9	8.2
HOL (%)	47.8	4.63	44.1	63.5
$\alpha$ -CEL (%)	40.7	3.02	-	38.1
KLAS (%)	7.9	0.65	3.8	25.1
Source	This work	Chao <i>et al.</i> 1999	Sharma <i>et al.</i> 2018	Jiménez <i>et al.</i> 2007

The low level of Klason-positive (acid insoluble) compounds detected was remarkable in comparison with agricultural residues or wood sources, which are usually around 20% to 30% in lignin (Jiménez *et al.* 2008), and very similar to the amounts reported



for *U. prolifera*, 9.4% (Li *et al.* 2016), or *U. reticulata* (Yoza and Masutani 2013). While Li *et al.* (2016) referred to the acid insoluble material as lignin, it is unlikely that algae could contain such high amounts of that heteropolymer, as they generally lack lignified cell walls (Martone *et al.* 2009). In any case, the low amount of hydrolysis-resistant compound makes *Ulva* a structurally fragile material (Lee *et al.* 2014) and therefore easily degradable during pulping, requiring much less energy and alkali charge than softwood, and less than hardwoods and common agricultural residues (Jimenez *et al.* 2008; Sharma *et al.* 2018).

### Characterization of Paper Sheets

Table 3 shows the values of the properties of the handsheets obtained after alkaline and hydrogen peroxide treatments and refining of the algal biomass. The average standard deviation (SD) is reported, showing low random errors for all tests. The highest value of tear index obtained (30.48 mN m<sup>2</sup>/g) was elevated compared with the paper sheets formed with bleached hardwood and even some softwood kraft pulps, and it was also high when it was compared with a non-wood raw material such as cotton stalks or vine shoots (Jiménez *et al.* 2007). Moreover, burst resistance was higher than many other pulps from vegetal residues such as rice straw, palm oil tree, or *Leucaena* (López *et al.* 2008; Rodríguez *et al.* 2008; Alriols *et al.* 2009). These results were in concordance with other reports where the mechanical properties of materials drastically benefited from the combination of conventional fibers and algal biomass (Seo *et al.* 2010; Khalil *et al.* 2017).

**Table 3.** Physical Properties of the Paper Sheets (50% *Ulva sp.* / 50% *Pinus pinaster*)

Assay	TI (mN m <sup>2</sup> /g)	BL (m)	ST (%)	BR (kN/g)	B (%)
1	24.54	2851	2.14	2.11	46.2
2	29.32	2999	2.21	2.57	58.4
3	26.14	2950	2.18	2.40	57.0
4	22.59	2536	1.95	1.77	53.3
5	19.94	2439	1.86	1.60	58.4
6	30.48	3163	2.30	2.66	40.2
7	28.35	2975	2.20	2.46	37.4
8	22.39	2549	1.97	1.92	40.2
9	19.85	2486	1.88	1.65	37.6
10	23.59	2830	2.12	2.06	54.1
11	24.96	2866	2.17	2.21	39.2
12	27.31	2983	2.20	2.51	46.3
13	20.33	2515	1.92	1.73	46.3
14	25.72	2892	2.16	2.34	47.6
15	23.62	2834	2.13	2.04	48.4
TI: tear index (SD = 0.63 mN m <sup>2</sup> /g), BL: breaking length (SD = 2 m), ST: stretch, BR: burst index (SD = 0.80 kN/g), B: brightness (SD = 0.58%).					

Chao *et al.* (2000) found that handsheets made from the red algae *Rhizoclonium* generally lacked bursting, tearing, and folding strengths (due to the morphological characteristics of the algal strands), but the proper combination with softwood pulp increased the tensile breaking length, tearing index, burst index, and the folding endurance at levels comparable to a typical kraft pulp. More interestingly, the addition of algal biomass from the municipal wastewater treatment to a conventional paper pulp enhances

the mechanical strength of the latter (Ververis *et al.* 2007). The increase of breaking length is attributed to the elevated proportion of proteins and its combination with chitin, which acts as a natural binding agent. The results in this work agree with other studies reporting a loss of whiteness (Ververis *et al.* 2007) as the complete removal of chlorophylls, which show high absorbance of blue and red light even at very small concentrations, is unfeasible.

All experimental data presented in Table 3 were fitted to a second-degree polynomial model as described above. Equations 4 through 8 were attained. The regression considered the five independent variables. Statistically significant terms of the polynomial model were selected with the double standard of having a value of Snedecor's F and Student's t higher than 4 and 2 respectively. These statistical results, together with the values of  $R^2$  (Eq. 4-8) being close to 1, establish a good functional relationship between input and output model parameters. The insignificant lack-of-fit implies that the regression models are well fit (Anupam *et al.* 2018).

$$TI = 24.146 + 1.26 X_S - 0.445 X_P + 3.65 X_R - 0.458 X_{PR} + 0.694 X_S^2 \quad (4)$$

$$R = 0.997$$

$$BL = 2854.689 + 45.476 X_S - 28.581 X_P + 254.433 X_R - 95.073 X_R^2 \quad (5)$$

$$R = 0.993$$

$$ST = 2.144 + 0.151 X_R - 0.077 X_R^2 \quad (6)$$

$$R = 0.964$$

$$BR = 2.135 + 0.111 X_S - 0.005 X_P + 0.393 X_R \quad (7)$$

$$R = 0.995$$

$$B = 47.375 + 8.663 X_P - 1.145 X_{SP} \quad (8)$$

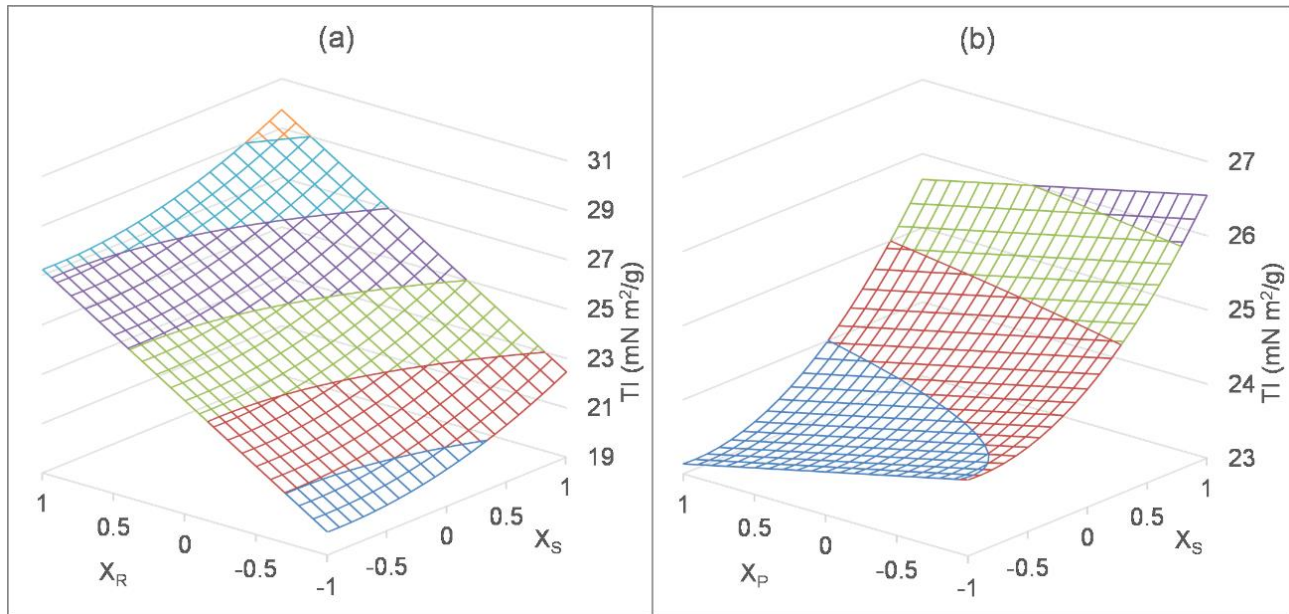
$$R = 0.984$$

In Eq. 4, the coefficients of  $X_S$  and  $X_R$  have a greater influence on the tear index, that is, it gradually increases with an increase of the refining and soda variables. However, it is clearly more influenced by refining (Fig. 1a), probably due to a good hydration of the fibers (more flexible after refining) resulting in its swelling, increasing its specific surface and therefore the contact points (Seth 1999; Hubbe *et al.* 2007). With respect to the hydrogen peroxide concentration (Fig. 1b), the influence is barely noticeable. Similarly, the *Brassica napus* pulps studied by Aguado *et al.* (2015) and Moral *et al.* (2017) showed gradual variations in the tear rate when revolutions of the PFI mill were increased.

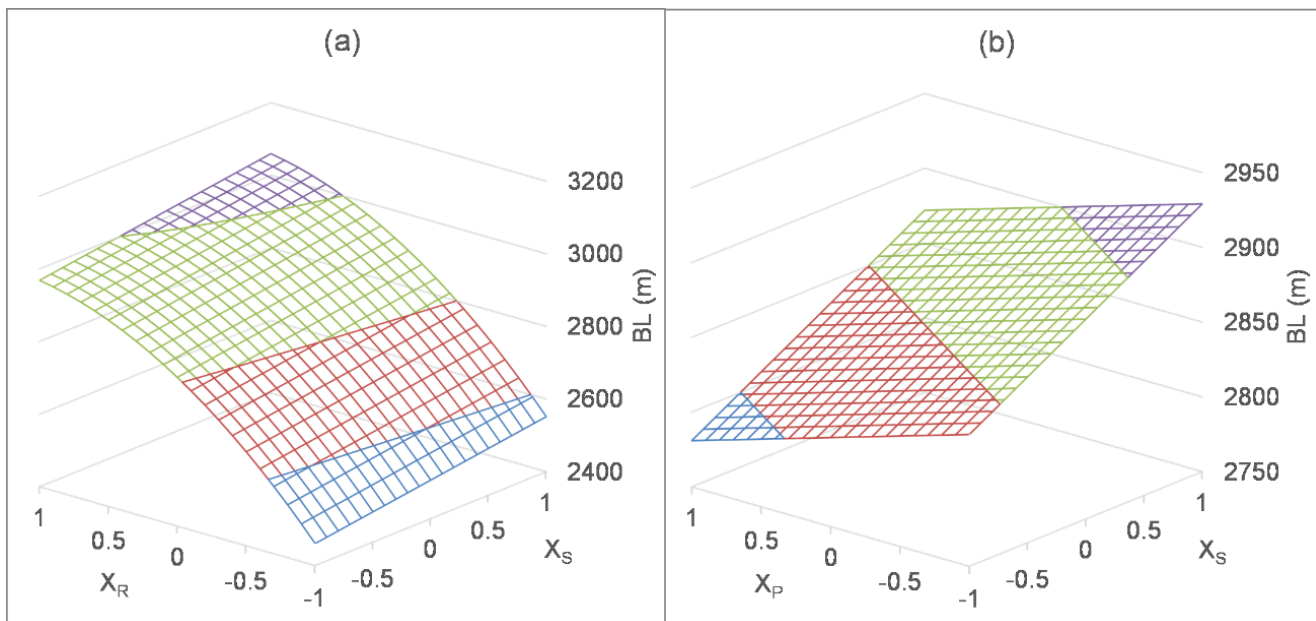
The second-degree polynomial equation obtained by regression (Eq. 5) reveals that the response variable breaking length was also more influenced by refining than by the concentration of soda. This may be due to the preservation of the natural fiber strength during pulping, as potential weak points in the sheets were minimized by a good removal of non-fibrous elements with little damage to carbohydrates. Then, the internal and external fibrillation caused by refining resulted in higher tensile resistance, without the drawbacks of fiber shortening that hardwoods usually suffer (Fig. 2).

Stretch was only influenced by refining (Eq. 6), and there was an increase in this response variable and elongation, reaching its maximum value (2.3%, Table 2) at the maximum value of PFI revolutions (5000). However, burst index was not only influenced by refining but also by the amount of soda, which had a remarkable contribution (Eq. 7). Fiber length and the generation of bonds between the fibers (pine and *Ulva* sp.) play a strong hand in this type of test, which suggests a larger contact surface in addition to a

greater number of interfibril interactions with the increase of the number of revolutions. Finally, as expected, peroxide concentration was the parameter that impacted whiteness the most (Eq. 8) due to the generation of species responsible for the oxidation of the chromophores (Abdel-Halim and Al-Deyab 2013).



**Fig. 1.** Response surfaces for the tear index with a) refining and soda concentration; b) hydrogen peroxide and soda concentration; in any case, the normalized values of other independent variables are set at 0



**Fig. 2.** Response surfaces for the breaking length with a) refining and concentration of soda; b) concentration of peroxide and concentration of soda; in any case, the normalized values of other independent variables are set at 0



## CONCLUSIONS

1. Chemical characterization of residues of *Ulva sp.* revealed high abundance of holocellulose (47.8%), mainly  $\alpha$ -cellulose, while the percentage of acid-insoluble material (7.9%) was low in comparison with conventional, lignin-containing fiber sources.
2. This green seaweed was deemed a promising raw material for its use in papermaking due to its ease of processing, consequently needing a relatively short pulping time, low consumption of energy and low concentration of chemical reagents (90 °C, 60 min, and 0.06 g to 0.1 g of NaOH per g of material).
3. Pulps from *Ulva sp.* combined with *Pinus pinaster* provided paper sheets with good physical properties, even surpassing the tear resistance of softwood fibers alone, and thus they should be considered when in search of new alternative sources of fibers for papermaking.

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