

Comparison of Natural and Synthetic Sorbents' Efficiency at Oil Spill Removal

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The spill of crude oil products into the environment has a negative impact on the ecosystem. Sorption materials are utilized as the means of their elimination. The sorption capacity of selected organic and inorganic natural sorbents, such as needles (*Larix decidua*, *Abies alba*, and *Pinus sylvestris*), sawdust from logging (*Fagus sylvatica*, *Picea abies*), leaf residues (*Fagus sylvatica*), moss (*Ceratodon purpureus*), soil, and synthetic sorbents Absodan Plus, expanded perlite, Eco-dry plus, and Reo Amos were all tested according to the standard ASTM F726 (2012). The natural sorbents were tested at various moisture contents (wet, air-dry, and dry) ranging from 0 to 82%. The pollutant used in the experiment was the low-viscosity engine oil 10W 40. The best sorption capacity among the wet sorbents was achieved with larch needles (11.1 g/g). Moss exhibited the best sorption capacity (25.2 g/g) among the air-dry sorbents. Regarding air-dry sorbents, larch needles, spruce sawdust, and beech sawdust showed the best results. When further dried, their sorption capacity decreased. Soil was the least efficient natural sorbent with a sorption capacity that ranged from 0.45 to 3.82 g/g. The best sorption capacity of 11.5 g/g among the synthetic sorbents was in Reo Amos. The sorption capacity of natural and synthetic substances was comparable.

Keywords: Natural sorbents; Synthetic sorbents; Oil spills; Sorption capacity

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INTRODUCTION

Despite the constantly growing utilization of biomass and other alternative energy sources and substituents, petroleum or crude oil products are still being used to a considerable extent. The most common crude oil products include oils, lubricants, fuels, such as diesel fuel or petrol, Vaseline, engine oils, and gearbox oils. The advantageous properties behind their usage are their ability to decrease friction, corrosion protection, heat transfer under mechanical friction, and use as fuels in combustion engines and spark-ignition engines. There are situations, including transportation, processing, pumping, careless handling and using or even an accident, when a spill or spilling in the environment can occur. When there is a spill from a transportation device or from vehicles, these substances spread across the solid surface or across the surface of liquids such as flowing or still waters. Another disadvantage of almost all kinds of oils, lubricants, and fuels, is their flammability, and in the case of an accident, their negative impact on the environment. Crude oil and crude oil products are classified as hazardous substances. For example, car petrol is classified by Regulation EC No.1272/2008 as flammable liquid substances (Flam.

Liq. 1), skin irritating (Skin Irrit. 2), and toxic inhalation (Asp. Tox. 1). It can cause cancer (Carc. 1B) and genetic damage (Muta. 1B) and is suspected of reproduction toxicity (Repr. 2). Oil spills also present a considerably negative impact on the environment; moreover, they are toxic to water organisms (Aquatic Chronic 2). Therefore, it is crucial in the case of a spill to eliminate their spreading and to decontaminate the affected area, soil, or water surface. There are various sorption materials used as prevention and also intervention measures.

Synthetic sorbents were not originally intended as sorption materials, but rather as waste material or a side product of industrial processing. Ash and pet dust are examples of products not originally intended as sorbents for spilled oil (Sun *et al.* 2013). Their ability to take up oil-based substances is seen as a side effect that led to the beginning of their use. The examples of synthetic sorbents include ash, other rock-based substances, like perlite, clay, and limestone (Rajaković-Ognjanović *et al.* 2008; Cheng *et al.* 2011; Bi *et al.* 2013; Kończewicz *et al.* 2013; Song *et al.* 2014; Wang *et al.* 2014; Bandura *et al.* 2015), plastics—polypropylene and polyurethane (Choi *et al.* 2011; Lee *et al.* 2013; Ge *et al.* 2014; Saleem *et al.* 2015; Zhang *et al.* 2017), carbon (Dong *et al.* 2012; Yoon *et al.* 2014), cellulose (Feng *et al.* 2015), and others. Synthetic sorbents are intentionally manufactured to serve as sorption materials to remove carbon and chemical spills. They might be further treated, for example by hydrophobization, which makes them hydrophobic and oleophilic. These types of sorbents have been processed as high-quality sorption materials.

For a spill into the environment, various natural sorption materials are used, which can be found naturally in the surroundings of the spill or the incident. and can be used instantly or at least until other sorbents are available. Another advantageous property of natural sorbents is their biological degradability. Some types of the sorption materials can affect the sorbed substance by various biological reactions and may degrade the substance being sorbed (Gertler *et al.* 2009; Lin *et al.* 2014; Olalekan *et al.* 2014). The use of inorganic and organic natural sorbents to remove oil products has been investigated (Singh *et al.* 2014; Wong *et al.* 2016). Natural sorbents include wood sawdust (Annunciado *et al.* 2005; Očkajová *et al.* 2016; Očkajová *et al.* 2018; Meng *et al.* 2019), cellulose-based material (Hubbe *et al.* 2013), natural clay (Zadaka-Amir *et al.* 2013), leaves (Sidik *et al.* 2012), rice straw (Hoang *et al.* 2018a; Hoang *et al.* 2018b) peat (Ribeiro *et al.* 2003), cotton (Carmody *et al.* 2007) and various natural fibres (Moriwakia *et al.* 2009; Abdelwahab 2014; Idris *et al.* 2014).

There are some sorption materials, either natural or synthetic, that can be used repeatedly (Karakutuk and Okay 2010; Korhonen *et al.* 2011; Bi *et al.* 2012; Hashim *et al.* 2012; Sun *et al.* 2013; Gu *et al.* 2014; Pham and Dickerson 2014; Qiu *et al.* 2015).

Sorbents generally can be described as insoluble substances or compound substances most often of solid state intended to capture spilled substances in a liquid or gaseous state. The sorption capacity depends on the kind of oil product, the size of the surface area where the oil product can be sorbed, and the type of surface (Fingas 2016). The surface size of sorption materials and increasing their sorption capacity depends on the amount of inner cracks, pores, and capillaries, as shown in Fig. 1.

A typical feature of sorption materials is their ability to sorb a certain amount of liquid in relation to their own weight (Ankowski *et al.* 2011). The sorbents appear mostly in the form of textile or a loose material. The loose sorption materials can further be divided into synthetic and natural. Currently, a majority of commercially available sorption materials are synthetic materials and they are of a hydrophobic nature, which makes them a perfect sorption material for various oil-based substances (Wang *et al.* 2013).

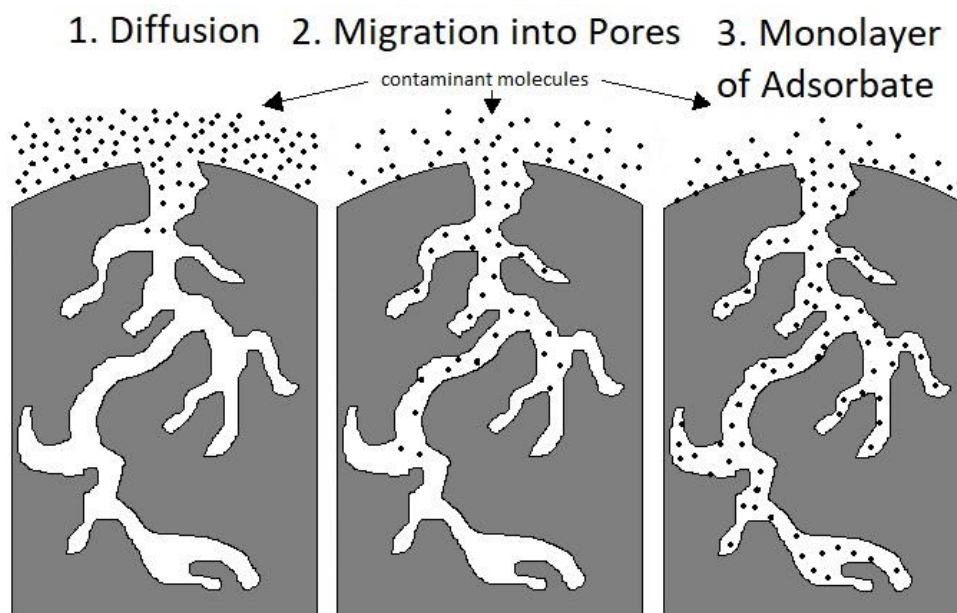


Fig. 1. Adsorption mechanism (redrawn from Vivek 2012)

When there is a spill of the hazardous material into the environment, it is important to stop the spill, limit its spread, and consequently facilitate its removal and disposal. A suitable way to decontaminate such a polluted environment is the utilization of various sorption materials that are often used at emergency services interventions.

The aim of the paper is to compare the sorption capacity of organic and inorganic natural sorption materials for used low-viscosity engine oil 10W 40 according to ASTM F726 (2012). The natural sorbents were chosen from an area that is prone to the spill of oil products because of the presence of heavy traffic roads and frequent occurrence of car accidents. This paper focuses on the spill of low-viscosity oil only on solid surface found within the forest environments. It does not discuss the spill of low-viscosity oil onto water surfaces. The natural sorbents were tested at three different moisture contents (wet, air-dry, and dry) as the moisture content in nature changes depending on vegetation season and climatic conditions. Synthetic sorbents were chosen based on their frequent use by emergency services. The 10W 40 low-viscosity oil is a component in both combustion and spark-ignition engines.

EXPERIMENTAL

Materials

The samples of various loose materials taken for further testing were taken directly from the natural environment in the original condition while maintaining their shape and form. The samples were taken from mixed forests nearby the town of Zvolen (Slovakia) at 300 m above the sea level in the spring time of the year 2018. The samples from the surface included fallen needles, leaves (from the previous autumn), moss (from the prior year), surface soil (maximum 2 cm depth), and sawdust recovered immediately after felling the trees. The samples of natural sorbents taken in such fashion were tested at three different moisture contents ranging from 0 to 82%.

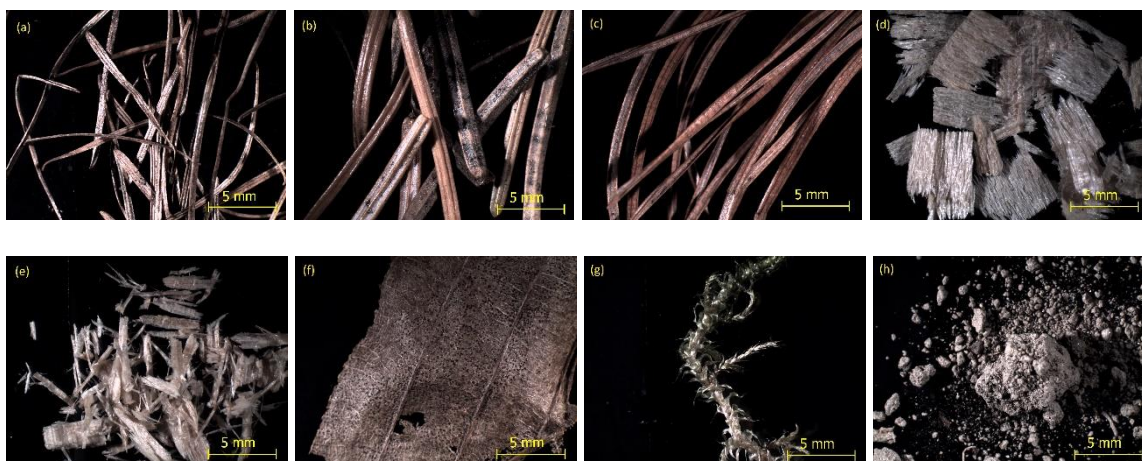


Fig. 2. Air-dry samples of natural sorbents: a) Larch needles, b) Fir needles, c) Pine needles, d) Beech sawdust, e) Spruce sawdust, f) Leaf residues, g) Moss, and h) Soil

Larch needles (*Larix decidua*), at the collection of samples, originally had a moisture content of 45.5% and the average needle size was $1 \times 1 \times 35 \text{ mm}^3$. Fir needles (*Abies alba*) originally had a moisture content of 27.2% and the needle size was $1 \times 2 \times 30 \text{ mm}^3$. Pine needles' (*Pinus sylvestris*) moisture content was 26.8% and the needles were $1 \times 1 \times 75 \text{ mm}$ in size. Beech sawdust (*Fagus sylvatica*) had a moisture content at sampling of 33.9% and the particle size was $0.5 \times 4 \times 4 \text{ mm}^3$. Spruce sawdust's (*Picea abies*) moisture content was 46.2% and the size of particles was $0.5 \times 2 \times 4 \text{ mm}^3$. The leaf residues' (*Fagus sylvatica*) moisture content was 34.2% and the particle size was $0.3 \times 70 \times 40 \text{ mm}^3$. Moss' (*Ceratodon purpureus*) moisture content at sampling was 81.9% and the particle size was $3 \times 3 \times 50 \text{ mm}^3$. Lastly, the soil's moisture content was 26.4% and the particle size was from 0.1 mm up to 4 mm.

The first test was conducted immediately after sampling; the samples were marked as wet and their moisture content was determined. The second test was performed when the samples had been air-dried (marked as air-dry). They had been dried at the room temperature until the stable state. The third test was performed with the samples dried to the absolute moisture content of 0 to 2% at the 103° to 105 °C temperature range (marked as dry samples). The moisture content was determined *via* a gravimetric method by universally applicable lab heating/drying oven with air circulation (UNB 200; Memmert Ltd., Schwabach, Germany). Table 1 states the mentioned tested sorbents and their average moisture content at each measurement.

The synthetic sorption materials commonly used during interventions were chosen for comparison. The synthetic sorbents were Absodan Plus, expanded perlite, Eco-dry Plus, and Reo Amos. The synthetic sorbents were in the loose form in an untreated condition as shown in Fig. 3. Absodan Plus (Imerys Industrial Minerals Denmark A/S, Fur, Denmark) is a mineral-based loose sorbent, based on Danish Moler clay 4 with the particle size of 1.5 to 3 mm. It is suitable to sorb liquids from a hard surface. Absodan Plus is a versatile sorbent for almost all types of liquids such as oils, acids, alkaline substances, oil products, water, and organic solvents.

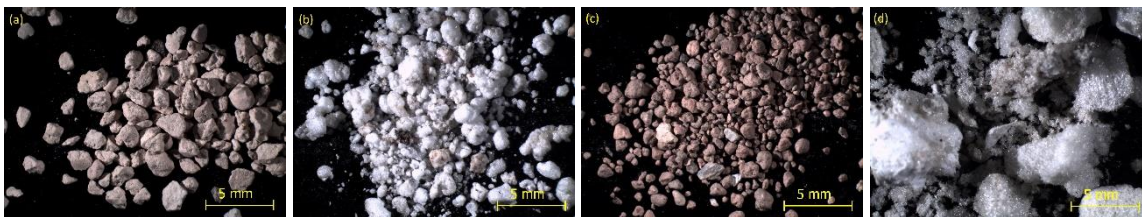


Fig. 3. Samples of synthetic sorbents: a) Absodan Plus, b) Expanded perlite, c) Eco-dry Plus, and d) Reo Amos

Expanded perlite (LBK Perlite Ltd., Lehôtka pod Brehmi, Slovakia) is an expanded amorphous aluminous silicate that is of a volcanic rock origin in a grain form with particles 0.5 to 2.5 mm in size. It has excellent thermal insulation and sound insulation properties. It is non-flammable and antimicrobial resistant and mold resistant. It is most often used to clean garages, pump stations, floors of production facilities, and other facilities where a spill of oil products could occur. Eco-dry Plus (REO AMOS SLOVAKIA, Ltd., Bratislava, Slovakia) is a non-dust, non-flammable, and health safe, rock-based crumb with 0.5 to 2.5 mm sized particles. It is suitable for removing or treating oil products, cutting and cooling emulsions, and aqueous solutions that it can sorb rapidly. It is a chemically stable substance, excluding hydrofluoric acid and 50% sodium hydroxide. Reo Amos (REO AMOS SLOVAKIA, Ltd., Bratislava, Slovakia) is a slightly dusty mixture of loose sorbents, it is granulated, chemically stable, and treated by a hydrophobic treatment. The particle size is 0.5 to 5 mm and it is designed to remove oil products from various rugged surfaces and still waters.

Sorption properties of the selected sorbents were tested for sorption of an oil product- the used engine oil 10W 40 (Total Slovakia, Ltd., Bratislava, Slovakia), which is commonly used in combustion and spark-ignition engines. The engine oil was chosen because it spills most often during car accidents at road collisions.

Methods

The moisture content of the natural sorbents at the individual tests was determined gravimetrically by drying in a kiln at the temperature of 103° to 105 °C to a constant weight (Kačík and Solár 1999).

The sorption capacity was stated by the standard testing method of sorption properties of adsorbents ASTM F726 (2012). The mentioned testing method refers to the laboratory testing of sorbents' properties to remove oil substances and other floating, non-soluble liquids that do not form emulsions. The method was applied to test type II (loose) sorbents. The tested sorption materials were conditioned at 23 ± 4 °C and a relative humidity of $70\% \pm 20\%$ for 24 h before the actual testing as per ASTM F726 (2012). The natural sorbents were not conditioned for the wet and dry moisture content tests so that sorbents maintained their moisture content for testing. The samples of the sorption materials were tested according to the testing method for short-time sorption of oil substances in laboratory conditions. A total of 5.00 g of the sorbent was weighed for each test. The crystallizing dish (23-cm diameter) was filled with the test liquid at least up to 2.5 cm height, which equals to 1 L of the sorbate (the low-viscosity engine oil 10W 40) that was refilled to the original amount after every test. To achieve higher accuracy, the empty testing sample holder was left sunk for 1 min in the engine oil and then left to drop off for 15 min. This procedure enabled the authors to achieve more precise results as the differences in the oil absorbed in the mesh of the holder were eliminated. Every

measurement was repeated three times as stated in the standard. The achieved results were calculated according to Eqs. 1 and 2 and the average sorption capacity was calculated. The used sorbents and oils were ecologically disposed.

The sorbed crude oil substance in the sorbent m_1 (g) was calculated according to Eq. 1,

$$m_1 = m_2 - m_3 - m_4 \quad (1)$$

where m_1 is the weight of the sorbate (g), m_2 is the weight of the wet test sample holder, a crystallizing dish and wet sorbent (g), m_3 is the weight of the wet sample holder and the dish (g), and m_4 is the weight of the dry sorbent (g).

The sorption capacity a_1 (g/g) was calculated using Eq. 2,

$$a_1 = m_1 / m_4 \quad (2)$$

where a_1 is the sorption capacity (g/g), m_1 is the weight of the sorbate (g), and m_4 is the weight of the dry sorbent (g).

The results were evaluated in the STATISTICA program (Statsoft, Inc., version 7.0, Prague, Czech Republic). The average values of sorption capacities of the selected sorbents and the values of the moisture content were evaluated and interpreted by an analysis of variance and Duncan's test where the significance level is 5%.

RESULTS AND DISCUSSION

The research was based on the technical standard ASTM F726 (2012) focusing on the examination and comparison of the sorption capacity of several kinds of sorbents to remove engine oil spills. Natural sorbents were tested at three different moisture contents (wet, air-dry, and dry) that are stated in Table 1. The given figures demonstrate that natural sorbents occur in a wide range of moisture contents. This is due to the different ability of the natural materials to sorb water vapors from the air.

Table 1. Tested Natural Sorbents and Their Moisture Contents (%) at the Individual Tests

Sorbent	Larch Needles	Fir Needles	Pine Needles	Beech Sawdust	Spruce Sawdust	Leaf Residues	Moss	Soil
Wet (%)	45.53	27.23	26.75	33.86	46.22	34.19	81.94	26.45
Air-dry (%)	10.74	7.04	10.82	9.02	10.73	11.79	12.26	2.04
Dry (%)	0 to 2	0 to 2	0 to 2	0 to 2	0 to 2	0 to 2	0 to 2	0 to 2

The achieved values of the sorption capacities of the tested sorbents at different moisture contents and their statistical assessment is stated in Table 2.

The sorption capacity values for the low-viscosity engine oil 10W 40 differed widely and are given in Table 2 for the various moisture contents (0 to 82%, Table 1).

Table 2 shows the achieved average values of sorption capacity at various moisture contents including the statistical parameters (standard error, confidence interval, standard deviation, and relative standard deviation).

Table 2. Sorption Capacity of Natural Sorbents at Various Moisture Contents and Relevant Statistical Data

Sorbents	Moisture	Sorption Capacity				N	Standard Deviation	Relative Standard Deviation
		Average (g/g)	Standard Error (g/g)	95% Confidence Interval (g/g)				
Larch needles	Wet	11.10	0.78	7.74	14.46	3	1.35	12.2
	Air-dry	19.47	1.36	13.63	25.31	3	2.35	12.1
	Dry	4.45	0.33	3.02	5.88	3	0.58	12.9
Fir needles	Wet	5.48	0.22	4.52	6.43	3	0.38	7.0
	Air-dry	5.17	0.16	4.49	5.85	3	0.27	5.3
	Dry	4.59	0.10	4.15	5.02	3	0.17	3.8
Pine needles	Wet	5.99	0.51	3.79	8.20	3	0.89	14.8
	Air-dry	4.99	0.17	4.24	5.73	3	0.30	6.0
	Dry	4.45	0.33	3.02	5.88	3	0.58	12.9
Beech sawdust	Wet	5.26	0.35	3.75	6.77	3	0.61	11.5
	Air-dry	8.13	0.42	6.33	9.93	3	0.72	8.9
	Dry	7.01	0.06	6.75	7.26	3	0.10	1.5
Spruce sawdust	Wet	4.07	0.26	2.94	5.19	3	0.45	11.1
	Air-dry	7.75	0.57	5.29	10.20	3	0.99	12.8
	Dry	6.54	0.32	5.15	7.94	3	0.56	8.6
Leaf residues	Wet	9.29	0.38	7.67	10.91	3	0.65	7.0
	Air-dry	13.39	0.86	9.69	17.09	3	1.49	11.1
	Dry	15.07	0.43	13.21	16.93	3	0.75	5.0
Moss	Wet	7.48	0.43	5.64	9.31	3	0.74	9.9
	Air-dry	25.16	0.49	23.07	27.25	3	0.84	3.3
	Dry	28.45	0.72	25.37	31.53	3	1.24	4.4
Soil	Wet	0.81	0.04	0.63	0.99	3	0.07	8.8
	Air-dry	3.82	0.29	2.59	5.06	3	0.50	13.0
	Dry	0.45	0.04	0.30	0.61	3	0.06	13.8

For the wet test, when the moisture content of the natural sorbents was the highest, 26% to 82%, the best sorption capacity was achieved by larch needles with the value of 11.1 g/g. The lowest sorption capacity was achieved by soil with the value of 0.81 g/g. Table 2 shows that the increased moisture content (26.45%) of this natural sorbent negatively influenced its sorption capacity.

For the air-dry test of the natural sorbents when their moisture content ranged from 7 to 13% (except soil whose moisture content was 2%), the best sorption capacity was achieved by moss. Decreasing its moisture content from the original 81.9% to 12.3% caused a considerable increase of sorption capacity up to 25.2% (g/g). The change in moisture content positively influenced the sorption capacity of all natural sorbents, except fir needles and pine needles. The soil sorbent, which was minimally efficient during the wet test, increased its sorption capacity greatly, however, it remained minimally efficient compared to the other natural sorbents.

For the dry test, the decreased moisture content of natural sorbents led to an increase in the sorption capacity of moss and leaf residue samples. The most efficient was moss with 28.4 g/g, and the least efficient was soil, which was demonstrated not only in this test but in all tests at all moisture contents.

Despite the various moisture contents, the most influencing factor is the overall surface area. The sorption capacity for all tested samples was influenced most by the nominal surface area which is shown in detail in Figs. 2 and 3.

Figure 4 presents the variance of the sorption capacities of individual natural sorbents at three different moisture contents (wet, air-dry, and dry).

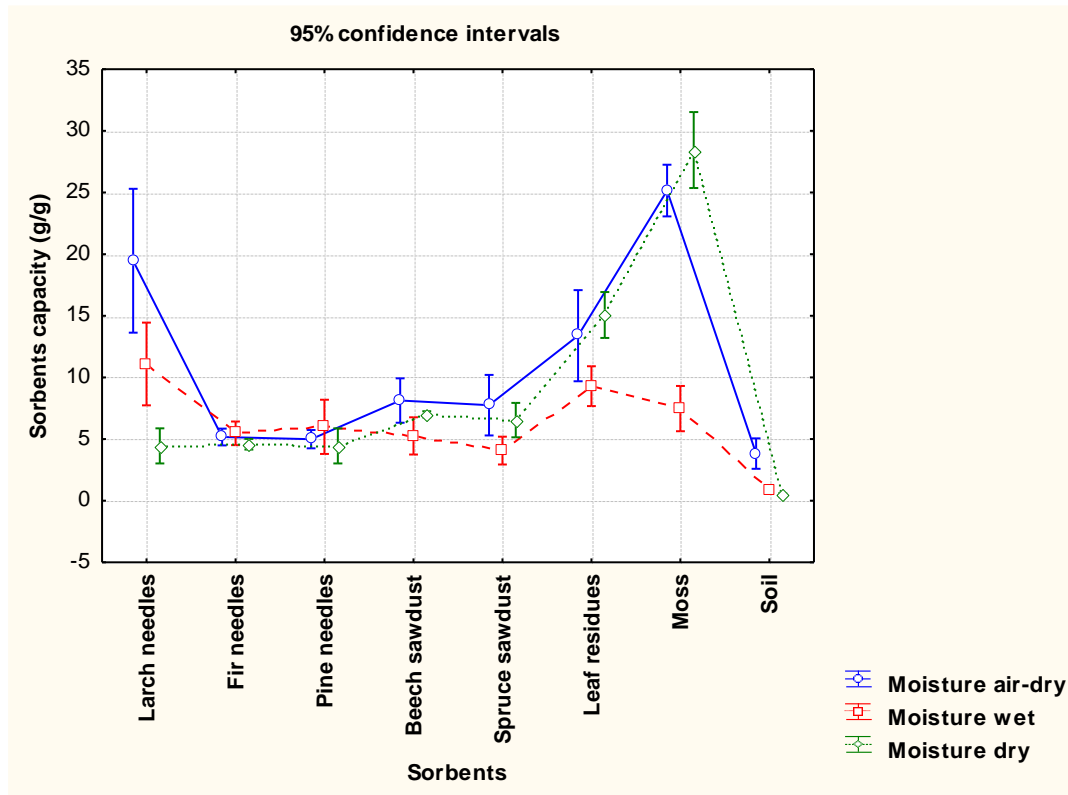


Fig. 4. Sorption capacity at different moisture contents

Comparing all three tests, it can be stated that the decreasing moisture content for the natural sorbent samples led to a slight decrease of sorption capacity for fir needles and pine needles; however, it implied an increase of sorption capacity for moss and leaf residues. For the other natural sorbents, the highest sorption capacity was found at the air-dry moisture content.

Tables 3, 4, and 5 state the Duncan test values for the chosen natural sorbents at different moisture contents. The results implied statistically significant differences at wet moisture content tests for larch needles, leaf residues, and soil, which state statistically significant differences in sorption capacity compared to all sorbents. The other tested sorbents stated non-significant statistical difference in some cases.

The results given in Table 4 show that a statistically significant difference at air-dry moisture content tests was achieved by larch needles, leaf residue, and moss, which

show statistically significant differences compared to all other sorbents. Other tested sorbents achieved non-significant statistical differences in some cases.

Table 3. Results of Duncan's Test for Wet Moisture Content Tests (P-values)

Moisture	Sorbents	Wet							
		1	2	3	4	5	6	7	8
Averages		11.10	5.48	5.99	5.26	4.07	9.29	7.48	0.81
Wet	1 Larch needles		0.000	0.000	0.000	0.000	0.014	0.000	0.000
	2 Fir needles	0.000		0.463	0.761	0.092	0.000	0.012	0.000
	3 Pine needles	0.000	0.463		0.331	0.022	0.000	0.059	0.000
	4 Beech sawdust	0.000	0.761	0.331		0.149	0.000	0.006	0.000
	5 Spruce sawdust	0.000	0.092	0.022	0.149		0.000	0.000	0.000
	6 Leaf residues	0.014	0.000	0.000	0.000	0.000		0.020	0.000
	7 Moss	0.000	0.012	0.059	0.006	0.000	0.020		0.000
	8 Soil	0.000	0.000	0.000	0.000	0.000	0.000	0.000	

Table 4. Table Results of Duncan's Test for Air-dry Moisture Content Tests (P-values)

Moisture	Sorbents	Air-dry							
		1	2	3	4	5	6	7	8
Averages		19.47	5.17	4.99	8.13	7.75	13.39	25.16	3.82
Air-dry	1 Larch needles		0.000	0.000	0.000	0.000	0.000	0.000	0.000
	2 Fir needles	0.000		0.800	0.000	0.002	0.000	0.000	0.104
	3 Pine needles	0.000	0.800		0.000	0.001	0.000	0.000	0.154
	4 Beech sawdust	0.000	0.000	0.000		0.593	0.000	0.000	0.000
	5 Spruce sawdust	0.000	0.002	0.001	0.593		0.000	0.000	0.000
	6 Leaf residues	0.000	0.000	0.000	0.000	0.000		0.000	0.000
	7 Moss	0.000	0.000	0.000	0.000	0.000	0.000		0.000
	8 Soil	0.000	0.104	0.154	0.000	0.000	0.000	0.000	

Table 5 presents the statistically significant difference for dry moisture content tests achieved by leaf residue, moss, and soil, which achieved statistically significant differences compared with all other sorbents. Other tested sorbents showed non-significant statistical differences in some cases.

Table 6 presents the maximum sorption capacity of the selected sorption materials of synthetic nature that were conditioned in laboratory conditions for 24 h before tests according to ASTM F726 (2012). The selected synthetic sorbents are commonly used in the interventions of emergency forces. As Table 3 states, the highest sorption capacity was found for Reo Amos with the sorption capacity of 9.24 g/g and the lowest for Absodan Plus with the sorption capacity of 1.06 g/g. An even lower sorption capacity of Absodan Plus for oils (0.5 to 0.6 g/g) is stated by Bandura *et al.* (2015), who tested sorption capacity of sorbents of the zeolite base. The sorption capacity was 3.33 g/g for expanded perlite,

which is probably the most frequently used sorption material of emergency forces interventions. Similar results for expanded perlite (2 to 4 g/g) are also stated by Teas *et al.* (2001).

Table 5. Results of Duncan's Test for Dry Moisture Content Tests (P-values)

Moisture	Sorbents	Dry								
		1	2	3	4	5	6	7	8	
Averages		4.45	4.59	4.45	7.01	6.54	15.07	28.45	0.45	
Dry	1	Larch needles		0.000	0.000	0.000	0.000	0.014	0.000	0.000
	2	Fir needles	0.000		0.463	0.761	0.092	0.000	0.012	0.000
	3	Pine needles	0.000	0.463		0.331	0.022	0.000	0.059	0.000
	4	Beech sawdust	0.000	0.761	0.331		0.149	0.000	0.006	0.000
	5	Spruce sawdust	0.000	0.092	0.022	0.149		0.000	0.000	0.000
	6	Leaf residues	0.014	0.000	0.000	0.000	0.000		0.020	0.000
	7	Moss	0.000	0.012	0.059	0.006	0.000	0.020		0.000
	8	Soil	0.000	0.000	0.000	0.000	0.000	0.000	0.000	

Table 6. The Sorption Capacity of the Selected Synthetic Sorbents

Synthetic Sorbents	Sorption Capacity	
	Average (g/g)	Variation \pm 15% (g/g)
Absodan Plus	1.06	0.16
Expanded perlite	3.33	0.50
Eco-dry Plus	1.37	0.21
Reo Amos	9.24	1.39

The comparison of natural and synthetic sorbents revealed that their sorption capacity is comparable; some natural sorbents displayed higher sorption capacity than synthetic sorbents, which is confirmed by other authors as well. For instance, Wong *et al.* (2016) present the overview of the sorption capacities of various natural materials and fibers and state that the application of natural sorbents based on plant fibers shows a great potential. Likon *et al.* (2013) compared the sorption capacity of various natural fibers and reported the outstanding sorption capacity of lignocellulose fiber made of *Populus nigra* 'Italica', whose sorption capacity for oil ranged depending on their density from 182 to 211 g/g.

Certainly, sorption capacity is not the only property of sorption materials that is considered during the selection of a proper sorbent. The other important properties include chemical resistance, non-flammability, cohesion, and the ability of mechanical loading of sorbents after sorption of a hazardous substance. Despite everything, the popularity of green materials (lignocellulose fibers, cellulose, modified cellulose, *etc.*) as sorption materials is constantly growing. They hold several advantages over synthetic sorbents. They are biodegradable, environmentally friendly, and do not cause secondary pollution (Chen *et al.* 2015). The further treatment of the used sorption materials is discussed in the Waste Oil Directive 75/439/EEC.

CONCLUSIONS

1. Regarding synthetic sorbents, the highest sorption capacity was achieved by Reo Amos at 9.24 g/g. The sorption capacity of synthetic and natural sorbents was comparable. The choice of suitable sorption material depends on a variety of factors, for instance, the kind of sorbed substance, chemical resistance of the sorbent, non-flammability, degradability, coherence, and mechanical loading, in addition to the price and availability.
2. When testing natural sorbents of the highest moisture content 26 to 82% (wet), the highest sorption capacity was demonstrated by larch needles at 11.1 g/g and the least efficient sorption capacity was shown by soil at 0.81 g/g. The increased moisture content of the natural materials negatively influences the sorption capacity, except the natural sorbents fir needles and pine needles.
3. During the air-dry tests of natural sorbents of the 2 to 13% moisture content, the highest sorption capacity was achieved by moss at 25.2 g/g. The samples of larch needles, spruce sawdust, and beech sawdust achieved the best results in this test; when they were further dried, their sorption capacity decreased.
4. When the moisture content further decreased (dry), the sorption capacity of moss and leaf residues increased. Moss was considered the most efficient sorbent of all tested sorbents. Its sorption capacity at this test was 28.4 g/g. The soil sample was the least efficient sorbent at all three moisture contents.
5. It can be stated that the parameter of moisture content of the tested natural sorbents is a statistically significant parameter towards its sorption capacity. The differences in the sorption capacity of eight tested natural sorbents were computed by Duncan's test. The test showed which values were statistically significant and they are shown in Tables 3, 4, and 5.
6. Subjected to an immediate use of the sorbent, during the spill of crude oil products in the environment, various natural sorption materials that can be found naturally in the surroundings of the spill can be used until another sorbent material arrives.

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