

Effect of Biomass Source on the Physico-mechanical Properties of Polyurethane Foam Produced by Microwave-assisted Liquefaction

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Different biomass sources (bamboo, rape straw, lignin, and Yaupon holly) were liquefied using microwave energy to produce biopolyols, which were then used to prepare biofoams without any further separation process. The results indicated that the content of hydroxyl groups in biopolyols derived from different biomass sources was sorted in descending order as rape straw, Yaupon holly, bamboo, and lignin. The rheological analysis demonstrated that the biopolyols were pseudoplastic, and the yield stress of biopolyols was remarkably increased with increasing biomass content. The compressive strength of polyurethane (PU) foam was rendered smaller by introducing biomass sources. Nevertheless, the biofoam obtained from biomass sources with higher hydroxyl groups content had better PU performance. In addition, the termite resistance performance of PU foam increased with the introduction of Yaupon holly, rape straw, and bamboo sources. Accordingly, the biofoams derived from the liquefaction of rape straw performed better than those from other biomass sources.

Keywords: Liquefaction; Polyurethane foam; Rheological analysis; Termite resistance

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INTRODUCTION

Polyurethane (PU) foam is one of the most versatile petroleum-based materials due to its low density, low thermal conductivity, and high mechanical performance (Zhang and Kessler 2015). It has been widely used in construction, packaging, and furniture production (Kausar 2017). With growing concerns for environmental damage and the rapid depletion of fossil fuels, massive efforts have focused on substituting petroleum-based polyols with bio-based ones, such as the biopolyol derived from biomass liquefaction (Huang *et al.* 2017a, 2018a).

Liquefaction is a promising way to convert biomass into valuable chemicals (Huang *et al.* 2017a). The main chemical components—namely, cellulose, hemicellulose, and lignin—in biomass can be broken down to lower molecular chemicals with numerous hydroxyl groups, such as C5 sugars, C6 sugars, and aromatics during the liquefaction process (Huang *et al.* 2017a). It has been demonstrated in previous works that the biopolyol produced by liquefaction is a promising feedstock with which to prepare bio-based PU

foam (Huang *et al.* 2017b, 2018b). Moreover, a considerable amount of biomass has been liquefied to prepare bio-based PU foams in different works, such as cornstalks (Yan *et al.* 2008), sugar cane bagasse (Hakim *et al.* 2011), wheat straw (Chen and Lu 2009), wood (Huang *et al.* 2018a), rape straw (Huang *et al.* 2018b), and lignin (Xue *et al.* 2015). Unfortunately, there are no comparisons among these works because the liquefaction parameters, including solid-liquid ratio, reaction conditions, catalyst, are different, resulting in a variability in the physico-mechanical properties of biopolyols (Huang *et al.* 2017b; da Silva *et al.* 2017). In addition, different amounts of bio-polyol substitution in the foaming process can cause the changes in bio-based foam properties.

Therefore, the liquefaction parameters were fixed to comparatively analyze the effect of selected biomass sources—specifically, bamboo, rape straw, lignin, and Yaupon holly—on the physico-mechanical properties of bio-based PU foams. The effect of the amounts of biopolyol substitution was investigated in this work as well. In particular, the microwave-assisted liquefaction products without any separation were used directly to produce bio-based PU foams. The characteristics of biopolyol were analyzed by Fourier transform infrared spectroscopy (FTIR) and a rheometer. The physico-mechanical properties, including the thermal stability, compressive strength, and termite resistance of the resulting biofoams, were evaluated in this work.

EXPERIMENTAL

Materials

Bamboo and rape straw were collected from the Sichuan province in China, and Yaupon holly was harvested at the Bob R. Idlewid Research Station near Clinton, LA, USA. Lignin was purchased from VWR International (Radnor, PA, USA). These raw materials were ground into 60- to 80-mesh particles using a Thomas Wiley Laboratory mill (Swedesboro, NJ, USA), and then oven-dried at 105 °C until constant weight.

Dibutyltin dilaurate (Pfaltz & Bauer, Waterbury, CT, USA) and 98% sulfuric acid (H₂SO₄) and was purchased from VWR International (Radnor, PA, USA). Materials kindly supplied by Huntsman Polyurethanes (Woodlands, TX, USA) were: polymeric methylene diphenyl diisocyanate (pMDI) (brand name: Rubinate M) with an average functionality of 2.7; isocyanate groups (NCO), content of 31.0% and viscosity of 192 cps at 25 °C; glycerol-based polyol with a hydroxyl value of 238 mg KOH/g (brand name: Jeffol FX 31-240); and a catalyst, dimethylcyclohexylamine (brand name: Jeffcat DMCHA) (Huntsman Corporation, Woodlands, TX, USA).

Dow Corning 193 (a silicone polyether surfactant from Dow Corning Corporation, Midland, MI, USA) and deionized water were used as surfactant and blowing agent, respectively. All chemicals were used without further purification.

Methods

Preparation of biopolyol

Liquefaction of bamboo, lignin, rape straw, and Yaupon holly were performed in a Milestone laboratory microwave oven (ETHOS EX, Shelton, CT, USA) equipped with an ATC-400FO automatic fiber optic temperature control system. A typical run for different biomass sources liquefaction (20% biopolyol substitution) was carried out with a loading of 2.4 g raw material, 12 g polyol, and 0.24 g sulfuric acid at 140 °C for 10 min (increasing

temperature for 5 min and maintained for 5 min). After liquefaction, the black products were used directly to produce bio-based PU foams.

Preparation of biofoams

The PU foams were prepared using a one-step method. A mixture of 3.00 g liquefaction mixtures, including polyol productions and unliquefied powders, 0.30 g co-catalyst (Jeffcat DMCHA: dibutyltindilaurate = 1:1), 0.20 g deionized water, and 0.20 g surfactant was thoroughly premixed with a mechanical stirrer for 1 min, followed by the addition of 10 g pMDI *via* stirring at 1500 rpm. The foams were allowed to freely rise in open plastic cylindrical cups and to cure overnight before testing. The PU foams with 0%, 10%, 20%, and 30% (by pure polyol) of Yaupon holly were labeled as PU0, PU10, PU20 (also named as PUY), and PU30, respectively. The foams containing 20% (by pure polyol) of different biomass sources (rape straw, bamboo, and lignin) were named as PUR, PUB, and PUL, respectively.

Chemical analysis of biopolyol

The FTIR spectrum of biopolyol was performed on a Nicolet Nexus670 spectrometer (Madison Instruments, Middleton, WI, USA) equipped with a Thermo Nicolet Golden Gate MKII Single Reflection ATR accessory. A small amount of sample was covered flatwise on the detection window. The scanning range of wavenumbers was from 4000 to 400 cm^{-1} with a resolution of 4 cm^{-1} . A total of 32 scans were collected.

Rheological properties of biopolyol

The rheological properties of biopolyol were measured using a stress-controlled rheometer of AR 2000 (AR-G2; TA Instruments, New Castle, DE, USA) equipped with a DIN concentric cylinder geometry. It consisted of a rotator with a diameter of 28.03 mm and a stainless-steel cup with a diameter of 30.38 mm. The rheological curves were obtained by measuring the viscosity or shear stress as a function of shear rate in the range of 0.1 to 1000 s^{-1} at 25 °C.

The rheological data of biopolyol were fitted to two different rheological models using Origin software (Originlab Corporation, v.2018C, Northampton, MA, USA). One of the most popular and efficient models to fit the relation between shear stress and shear rate is the Herschel-Bulkley model (Herschel and Bulkley 1926), which has the following form,

$$\tau = \tau_0 + K\dot{\gamma}^n \quad (1)$$

where τ is the shear stress (Pa), τ_0 is the yield stress (Pa), K is the flow consistency coefficient, $\dot{\gamma}$ is the shear rate (s^{-1}), and n is the flow behavior index. This model has been successfully used to describe the rheological properties of biopolyol obtained from the liquefaction of cellulose (Kosmela *et al.* 2016). Because this model cannot give a unique fit for a given data set, the rheological properties of biopolyol were also analyzed using an improved model, namely, the Sisko model (Sisko 1958), as given below,

$$\tau = \mu_\infty \dot{\gamma} + K\dot{\gamma}^n \quad (2)$$

where μ_∞ is the viscosity at infinite shear rate (s^{-1}), K is the flow consistency coefficient, and n is the flow behavior index. This model performed well in describing the rheological properties of complex fluids over the entire range of shear rate, for example, in drilling fluids (Li *et al.* 2016).

Mechanical properties of biofoams

An eXpert2610 universal mechanical test analyzer (ADMET, Norwood, MA, USA) was used to measure the compressive strength of the foams, according to the ASTM D695-10 (2010). Samples were placed between the two parallel plates and compressed at 10 mm/min. The Young's modulus was calculated by the slope of the tangent of linear portion in the stress-strain profile in accordance with the method described in previous reports (Gama *et al.* 2015). The compressive strength was taken from the stress-strain curves at a deformation of 20%. Ten replicates were measured for each group.

Chemical analysis of biofoams

The chemical analyses of PU foams were also characterized using FTIR.

Thermogravimetric analysis (TGA) of biofoams

Thermogravimetric analysis (TGA) with a thermal analyzer (Q50 TGA; TA Instruments, New Castle, DE, USA) was applied to obtain thermogravimetric data. Each sample of approximately 5 mg was conducted at 30 °C to 800 °C with a constant heating rate of 20 °C/min under a flow of 40 mL/min of nitrogen atmosphere.

Termite resistance of biofoams

Coptotermes formosanus (Brechtel State Park, Algiers, LA, USA) were used to test the termite resistance performance of biofoams in according to AWWPA E1-17 (American Wood Production Association Standard 2017). The test specimens were put into vessels and exposed to the termites for 4 weeks. Then, the weight loss was calculated to evaluate the termite resistance of biofoams.

Statistical analysis

Statistical analysis was carried out using SAS (version 9.1, SAS Institute, Cary, NC). The mean and the standard deviation were calculated, and analysis of variance (ANOVA) was performed to determine significant difference ($\alpha=0.05$) among the samples.

RESULTS AND DISCUSSION

FTIR Spectra of Biopolyols

Figure 1 presents the typical IR spectra of Yaupon holly biopolyols. For instance, the presence of hydroxyl groups was observed from peaks of 3450 cm^{-1} (Huang *et al.* 2017b). The characteristic absorbance was calibrated and normalized by referring to the baseline and phenyl band at 1597 cm^{-1} . When the content of Yaupon holly was increased from 10% to 20%, the intensity of hydroxyl groups increased. It was ascribed to the increased hydroxyl content in biopolyol derived from higher biomass liquefaction. However, by further increasing Yaupon holly content from 20% to 30%, the intensity of 3450 cm^{-1} decreased. This was because a higher biomass to solvent ratio resulted in a lower liquefaction yield (Huang *et al.* 2017b).

The absorbance band at 1730 cm^{-1} was due to either the vibration of the uronic ester and acetyl groups in hemicellulose, or the ester linkage of carboxylic groups of p-coumaric and ferulic in lignin (Flauzino Neto *et al.* 2013). It increased with increasing biomass content, which was attributed to a higher content of biomass being dissolved in biopolyol.

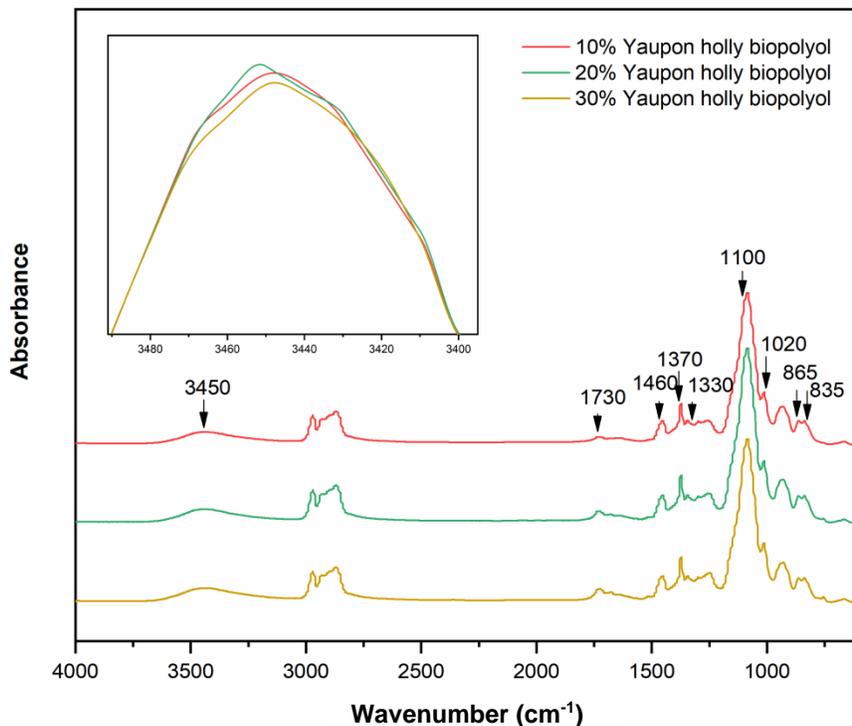


Fig. 1. FTIR spectra of Yaupon holly biopolyol *via* microwave liquefaction

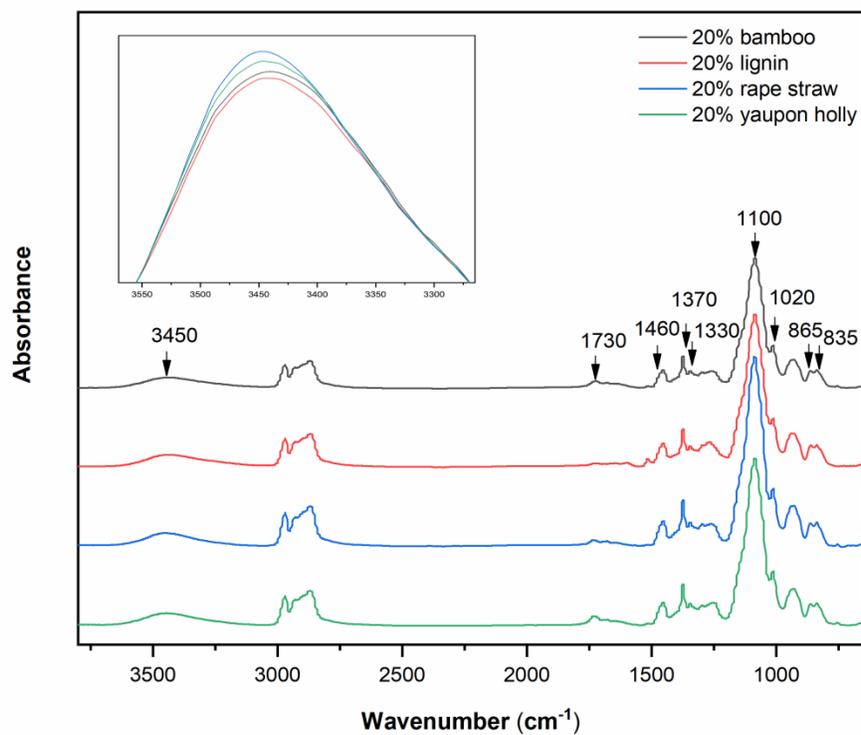


Fig. 2. FTIR spectra of biopolyol derived from different biomass sources *via* microwave liquefaction

The peak at 1460 cm^{-1} arose from C–H bending on the benzene ring (Alemdar and Sain 2008). The absorbance bands at 1370 cm^{-1} corresponded to phenolic OH groups (Huang *et al.* 2017b). The peaks at 1330 cm^{-1} were associated with the syringyl unit (Alemdar and Sain 2008). The prominent peaks at 865 cm^{-1} and 835 cm^{-1} were assigned to two adjacent hydrogen atoms on the benzene ring (Li *et al.* 2015). The C–O–C peak at 1020 cm^{-1} and 1100 cm^{-1} were assigned to the stretching vibration of C–O, and attributed to the degradation of cellulose (Alemdar and Sain 2008).

Figure 2 clearly shows the typical bio-based polyol FTIR spectra with hydroxyl groups' absorption peaks at 3450 cm^{-1} , cellulosic groups' absorbent bands at 1020 cm^{-1} , and lignin composition bands at 1370 cm^{-1} , *etc.* Because polyurethane foam from different biomass sources were compared in this work, the specific hydroxyl groups at 3450 cm^{-1} were sorted in descending order as follows: rape straw, Yaupon holly, bamboo, and lignin.

Rheological Properties of Biopolyols

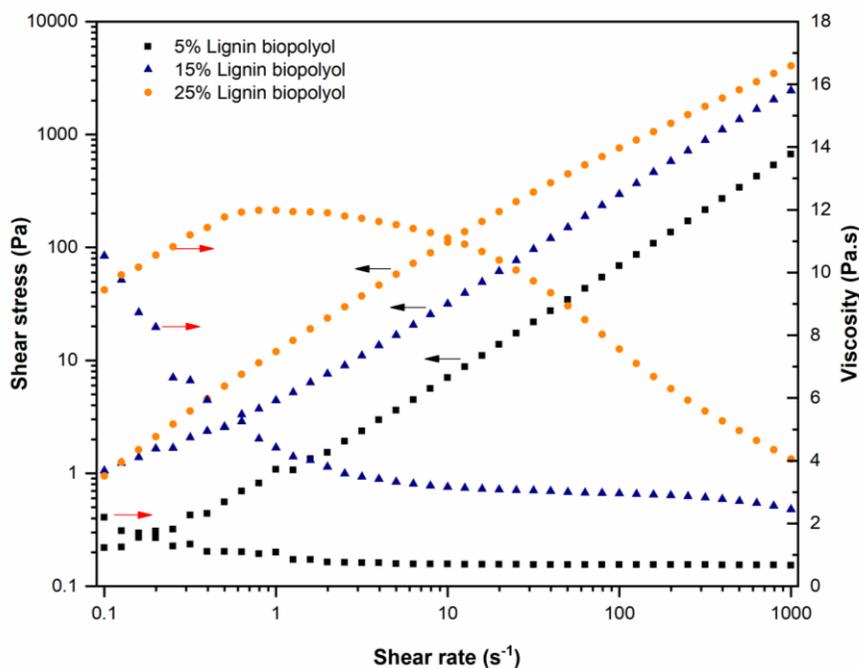


Fig. 3. Rheological properties of biopolyols with varying lignin contents

Lignin-based biopolyols were selected for analysis of the variation of rheological properties with increasing biomass content. Figure 3 shows the shear stress of biopolyols with respect to shear rate. The non-linear increase tendency of shear stress was observed in all the biopolyols with the increase of shear rate. A higher shear stress was observed when the biomass content increased, which was ascribed from the high viscosity, requiring higher shear stress to move. With an increasing shear rate, nonlinear reductions on the viscosity of biopolyols were found in Fig. 3. The decreasing tendency was attributed to the increase of liquid flow arrangement that resulted from increasing shear rate (Khalil and Jan 2012). The disproportionality of viscosity at low shear rates indicated that biopolyols can be categorized as non-Newtonian liquids (Li *et al.* 2016). The increasing viscosity corresponded to an increase in lignin content, which was attributed to more biomass dissolved with liquefaction.

The estimated mathematical models are shown in Table 1. Larger R^2 values were observed in the Sisko models, compared with in the Herschel-Bulkley models, confirming that the Sisko model was more appropriate for describing the rheological properties of biopolyol. However, the Herschel-Bulkley model provided the yield stress (intercept, τ_0) directly, indicating that the minimum shear stress was able to move the fluid. The yield stress remarkably increased when the lignin content increased from 5% to 25%, suggesting that it was difficult to move the fluid when the biocontent increased. Normally, the fluid could be defined as three types in the base of the flow index values. They types are, respectively, Newtonian ($n = 1$), pseudoplastic (non-Newtonian) with a shear thinning behavior ($n < 1$), and dilatant (non-Newtonian) with shear thickening behavior ($n > 1$) (Khalil and Jan 2012). The flow behavior indexes (n) obtained from the Herschel-Bulkley and Sisko models were lower than 1. This result indicated that the biopolyols were pseudoplastic fluids. It was confirmed that the pseudoplastic biopolyol was suitable for producing bio-based polyurethane foams (Khalil and Jan 2012).

Table 1. Rheological Models of Different Content of Lignin-based Biopolyols

Biopolyols	Herschel-Bulkley Model	Sisko Model
5% lignin	$y = 0.10769 + 0.74367x^{0.9846}$, $R^2 = 0.99999$	$y = -1.40488x + 2.14498x^{0.99509}$, $R^2 = 0.99999$
15% lignin	$y = 2.93162 + 5.12324x^{0.89571}$, $R^2 = 0.99947$	$y = -52.68097x + 57.06588x^{0.99509}$, $R^2 = 0.99963$
25% lignin	$y = 4.38339 + 26.1311x^{0.73164}$, $R^2 = 0.99968$	$y = -5.19652x + 23.64662x^{0.86408}$, $R^2 = 0.99994$

FTIR Spectra of Biofoams

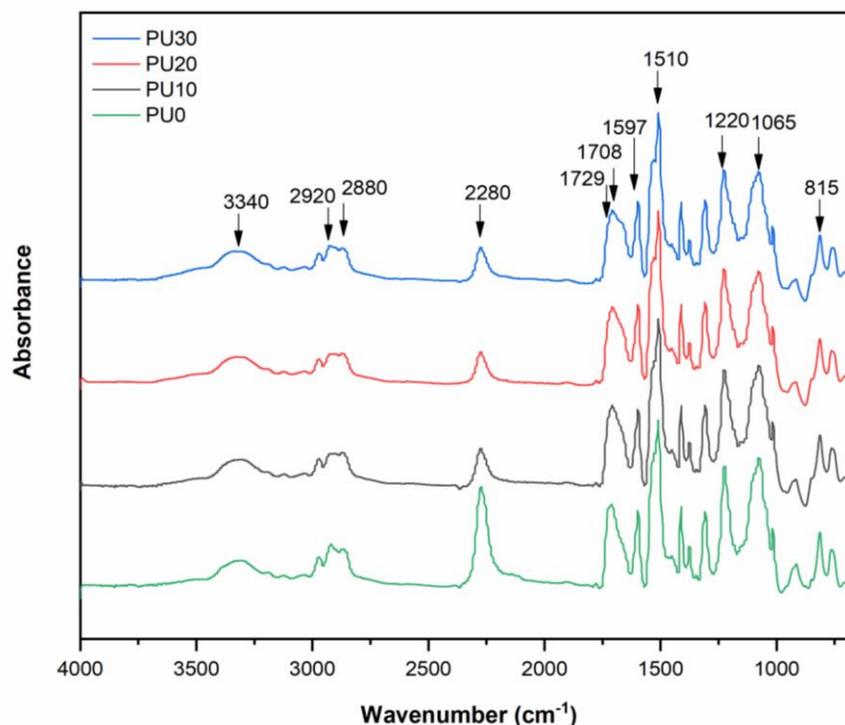


Fig. 4. FTIR spectra of polyurethane foams with varying Yaupon holly contents

The FTIR spectra of biofoams with different contents of Yaupon holly are shown in Fig. 4. The absorption peaks at 2920 cm^{-1} and 2880 cm^{-1} were derived from the C–H asymmetric and symmetric stretching vibration of $-\text{CH}_3$, respectively, indicating the presence of side chain structural units in lignin (Li *et al.* 2020). The peaks at 1597 cm^{-1} and 1510 cm^{-1} corresponded to the C = C stretching vibration of the benzene ring, which are characteristic peaks of lignin (Cheng *et al.* 2010).

The peak at 2280 cm^{-1} evidenced the existence of the NCO groups derived from isocyanate (Huang *et al.* 2018a). Its intensity decreased first when increasing the biocontent from 0 to 20%, and then increased as the biocontent increased to 30%. This result was associated with the increase in hydroxyl groups that could react with the NCO group, and then the decrease of liquefaction yield, resulting in the decrease of biopolyol production. The absorbance band at 1729 cm^{-1} corresponded to the vibration of the uronic ester and the acetyl groups in hemicellulose, or the ester linkage between carboxylic groups of p-coumaric and ferulic in lignin and hemicellulose (Li *et al.* 2015). The absorbance peak at 1708 in the carbonyl region was ascribed to the C = O stretching of ester groups (Ugarte *et al.* 2017). The absorption peak at 1220 cm^{-1} was the C–O vibration. The bands located at 815 cm^{-1} were assigned to the C–H bond of the out-of-plane bending vibration absorption peak on the benzene ring of lignin (Ugarte *et al.* 2017). These peaks were evidence that the liquefaction fragments were involved in the PU foam structure.

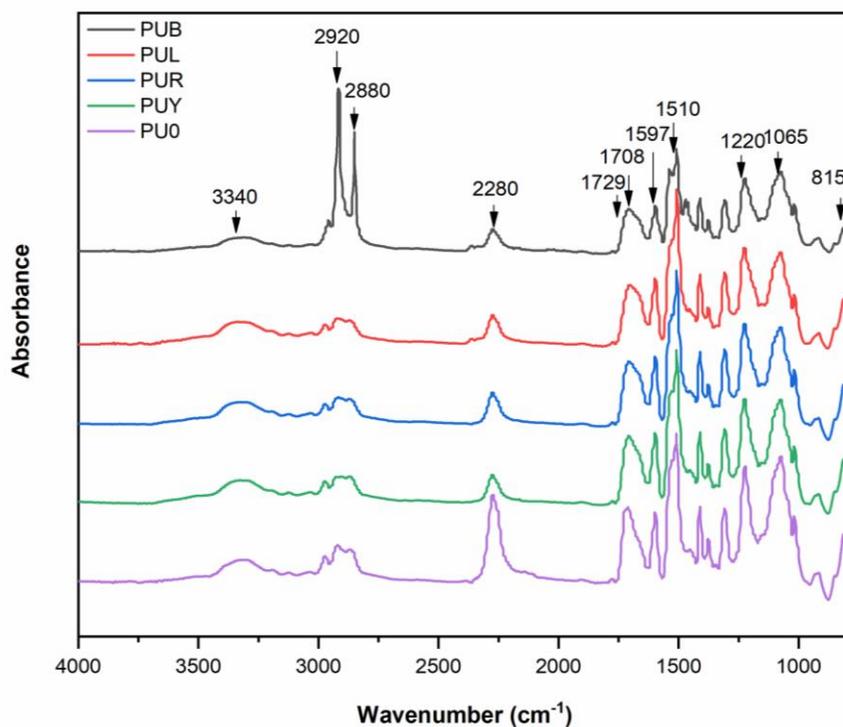


Fig. 5. FTIR of polyurethane foams derived from different biomass sources

The FTIR spectra of biofoams with different biomass sources are shown in Fig. 5. All FTIR spectra had similar shapes and intensities, which indicated that all biomass types had similar ways of participating in the foaming reaction. The intensities of NCO IR peaks of biofoams at 2280 cm^{-1} were weaker than those of the neat foam. This result further evidenced that the biomass could react with NCO groups. It was noteworthy that the FTIR

of foams with 20% bamboo biopolyol had obvious vibrations at the peaks of 2920 cm^{-1} and 2880 cm^{-1} . The major peak at 2920 cm^{-1} was derived from the C–H asymmetry, and the second one at 2880 cm^{-1} was caused by symmetric stretching vibration of $-\text{CH}_3$. This was because of the presence of the side chain structural units in lignin (Cheng *et al.* 2010).

Thermogravimetric (TG) and Derivative Thermogravimetric (DTG) Curves of Biofoams

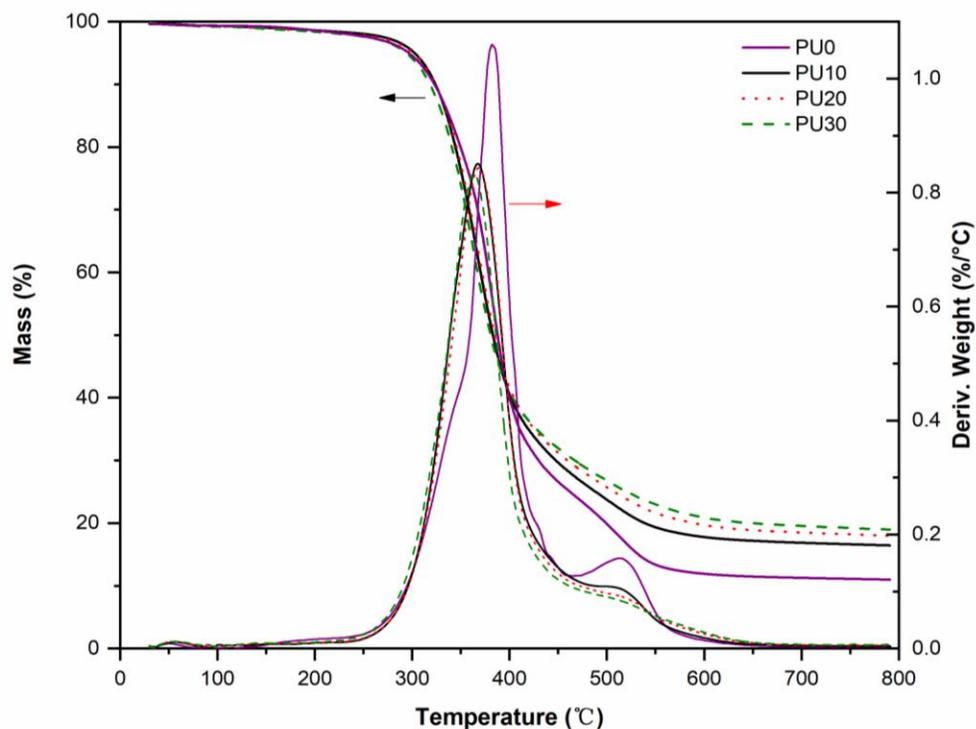


Fig. 6. TG and DTG curves of polyurethane foams with different Yaupon holly contents

The TG and DTG curves of biofoams and neat foams are shown in Fig. 6. There were three distinctive regions of major weight loss for these biofoams. The weight loss up to $100\text{ }^{\circ}\text{C}$ was considered to be due to the evaporation of moisture and the release of volatile components. The second degradation stage, up to $450\text{ }^{\circ}\text{C}$, was attributed to the decomposition of hemicellulose fragments, cellulose fragments, unstable urethane links, and rigid segments in polyurethane. The last region, up to $800\text{ }^{\circ}\text{C}$, was related to the pyrolysis of lignin fragments and the soft segments in foams (Huang *et al.* 2018a). The biochar residue increased with increasing biomass content. Lower maximum thermal degradation rates were found from biofoams, as compared with neat foam, at about $350\text{--}400\text{ }^{\circ}\text{C}$, which may be related to the higher thermal stability of lignin, crystalline region in cellulose and hemicellulose introduced from biomass sources (Delucis *et al.* 2018). No remarkable difference in the degradation temperature in the different contents of biomass was observed, except for the soft segments' pyrolysis at approximately $530\text{ }^{\circ}\text{C}$. The intensity of the pyrolysis of soft segments decreased with increasing biomass content, suggesting the involved biomass changed the polyurethane structure when the content of soft segments decreased. It was worthy of note that the PU0 presented a prominent peak at the region of $450\text{ to }550\text{ }^{\circ}\text{C}$. This phenomenon was probably due to the decomposition of the organic chain, which was mainly governed by cleavage of urea groups (Gu *et al.* 2013).

The thermal degradation properties of biofoams made from different biopolyol sources are shown in Fig. 7. All thermal degradation patterns of biofoams were similar, except for the degradation of soft segments at approximately 530 °C. The thermal stability of soft segments from rape-straw foam was lower than others, which might have been caused by the difference of chemical components among different biomass sources. The thermal stability of urethane links reflects the reaction density between isocyanate groups and hydroxyl groups. From Fig. 7, the degradation temperature of urethane links in biofoams at approximately 375 °C decreased in the order of rape straw, Yaupon holly, bamboo, and lignin, which was coincidental of the relationship of hydroxyl group IR intensities among different biomass sources. Furthermore, the decomposition temperatures of urethane links from biofoams were lower than that of neat form, which indicated that the introduction of biomass weakened the thermal stability of PU foams.

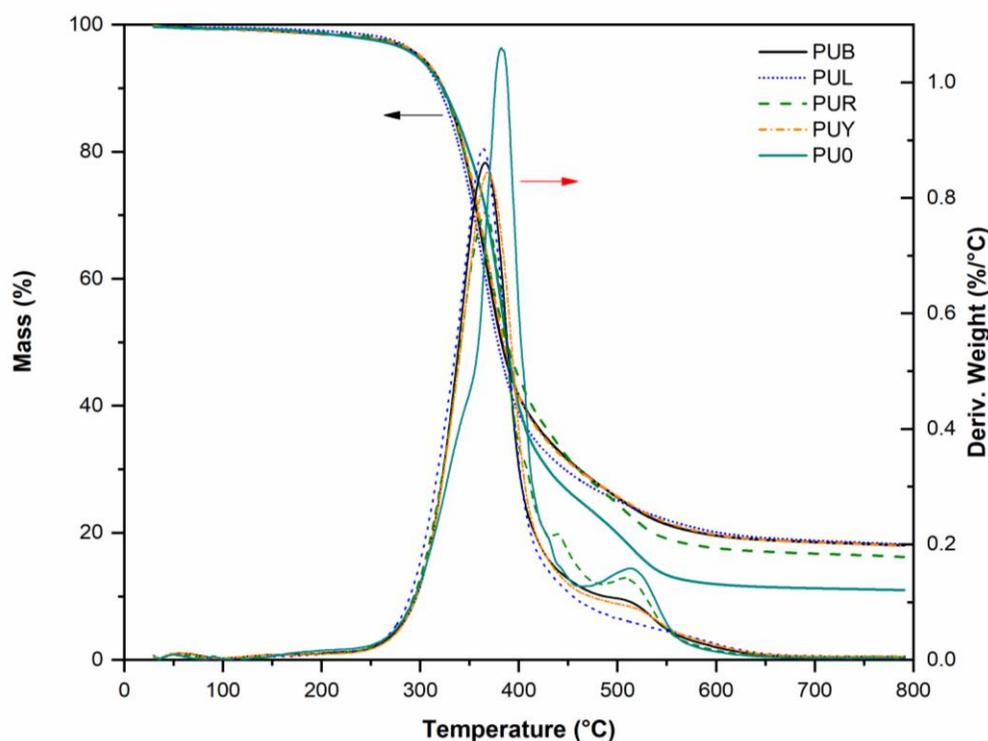


Fig. 7. TG and DTG curves of PU foams with different bio-polyol sources

Compressive Strength of Biofoams

The compressive strengths of biofoams at 20% strain with varying biocontents of Yaupon holly are shown in Table 2. The compressive strengths of the biofoams were lower than that of the neat foam. This was ascribed to low crosslinking density between biopolyol and isocyanate (Zhang *et al.* 2015). As for biofoams, no differences in the compressive strengths between PU10 and PU20 were observed. However, a slight decrease was found when the biomass content increased to 30%. This was attributed to a lower liquefaction yield that accompanied higher biocontent (Huang *et al.* 2017b). The compressive strength of biofoam with 20% rape straw was higher than other foams with 20% biomass. The compressive strength of biofoams was sorted as PUR > PUY > PUB > PUL. This tendency to change was attributed to the change in hydroxyl group content among different biomass sources.

Table 2. Compressive Strength at 20% Strain of Biofoams

PU Foams	Compressive Strength (δ 20% KPa)
PU0	148.0 \pm 11.8 ^a
PU10	53.0 \pm 7.8 ^c
PU20/PUY	53.0 \pm 9.9 ^c
PU30	40.8 \pm 9.4 ^c
PUR	81.6 \pm 12.6 ^b
PUL	46.9 \pm 4.7 ^c
PUB	51.4 \pm 3.7 ^c

Values with the same letters indicate there is no significant difference at 0.05 probability.

Termite Resistance of Biofoams

Table 3 presents the termite resistance performance of biofoams. The weight loss of the biofoams, except for lignin foam (PUL), was lower than that of PU0, which suggested that the neat foam matrix probably contain a higher content of nutrients needed by termites. The highest weight loss was observed in PUL, which showed that lignin was preferred by the termites (Subekti *et al.* 2015). When the biomass content increased from 10% to 30%, the weight loss increased slightly. This was attributed to the higher content of biomass providing a greater amount of food for the termites.

Table 3. Termite Resistance Performance of Biofoams

Foam Type	Weight Loss (%)
PU0	16.8 \pm 4.7 ^b
PU10	5.7 \pm 1.2 ^d
PU20/PUY	9.0 \pm 4.0 ^c
PU30	9.5 \pm 1.2 ^c
PUL	22.3 \pm 0.8 ^a
PUR	12.8 \pm 1.4 ^c
PUB	5.6 \pm 0.8 ^d

Values with the same letters indicate there is no significant difference at 0.05 probability.

CONCLUSIONS

1. The FTIR spectra of biopolyols demonstrated the successful liquefaction of biomass, and in biopolyols derived from varying biomass sources, the content of hydroxyl groups was sorted, in descending order, as rape straw, Yaupon holly, bamboo, and lignin.
2. Using rheological analysis, the biopolyols were determined to be pseudoplastic fluids. The yield stress of biopolyols increased remarkably with increasing biomass content.
3. Biofoams had a lower compressive strength than that of neat foam. Nevertheless, the biofoam obtained from biomass sources with higher hydroxyl groups content had better PU performance.
4. Rape straw, Yaupon holly, and bamboo sources in biofoams enhanced the termite resistance performance of PU foams. Lignin had the reverse effect.

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